

Spectral light absorption by ambient aerosols influenced by biomass burning in the Amazon Basin. I: Comparison and field calibration of absorption measurement techniques

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Abstract. Spectral aerosol light absorption is an important parameter for the assessment of the radiation budget of the atmosphere. Although on-line measurement techniques for aerosol light absorption, such as the Aethalometer and the Particle Soot Absorption Photometer (PSAP), have been available for two decades, they are limited in accuracy and spectral resolution because of the need to deposit the aerosol on a filter substrate before measurement. Recently, a 7-wavelength (λ) Aethalometer became commercially available, which covers the visible (VIS) to near-infrared (NIR) spectral range ($\lambda=450\text{--}950\text{ nm}$), and laboratory calibration studies improved the degree of confidence in these measurement techniques. However, the applicability of the laboratory calibration factors to ambient conditions has not been investigated thoroughly yet.

As part of the LBA-SMOCC (Large scale Biosphere atmosphere experiment in Amazonia – SMOke aerosols, Clouds, rainfall and Climate) campaign from September to November 2002 in the Amazon basin we performed an extensive field calibration of a 1- λ PSAP and a 7- λ Aethalometer utilizing a photoacoustic spectrometer (PAS, 532 nm) as reference device. Especially during the dry period of the campaign, the aerosol population was dominated by pyrogenic emissions. The most pronounced artifact of integrating-plate type attenuation techniques (e.g. Aethalometer, PSAP) is due to multiple scattering effects within the filter matrix. For

the PSAP, we essentially confirmed the laboratory calibration factor by Bond et al. (1999). On the other hand, for the Aethalometer we found a multiple scattering enhancement of 5.23 (or 4.55, if corrected for aerosol scattering), which is significantly larger than the factors previously reported (~ 2) for laboratory calibrations. While the exact reason for this discrepancy is unknown, the available data from the present and previous studies suggest aerosol mixing (internal versus external) as a likely cause. For Amazonian aerosol, we found no absorption enhancement due to hygroscopic particle growth in the relative humidity (RH) range between 40% and 80%. However, a substantial bias in PSAP sensitivity that correlated with both RH and temperature (T) was observed for $20\% < RH < 30\%$ and $24^\circ\text{C} < T < 26^\circ\text{C}$, respectively. In addition, both PSAP and Aethalometer demonstrated no sensitivity to gaseous adsorption. Although very similar in measurement principle, the PSAP and Aethalometer require markedly different correction factors, which is probably due to the different filter media used. Although on-site calibration of the PSAP and Aethalometer is advisable for best data quality, we recommend a set of "best practice" correction factors for ambient sampling based on the data from the present and previous studies. For this study, the estimated accuracies of the absorption coefficients determined by the PAS, PSAP and Aethalometer were 10, 15 and 20% (95% confidence level), respectively.

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1 Introduction

The Intergovernmental Panel on Climate Change (Penner et al., 2001) has identified radiative forcing by aerosols as one of the major uncertainties in the global radiation budget. While light scattered by aerosols cools the atmosphere (negative radiative forcing), absorbed electromagnetic radiation contributes to a positive radiative forcing. In addition to the direct heating of the atmosphere due to light absorption, there is a semi-direct effect as a result of the enhanced dissipation of clouds in the vicinity of heated aerosol layers (Ackerman et al., 2000; Penner et al., 2001). The latter may have significant implications on regional and global precipitation patterns. Despite its significance, light absorption by atmospheric aerosol is relatively poorly characterized in part due to a lack of reliable instrumentation.

Aerosol light scattering and absorption can be characterized by the scattering and absorption coefficients, σ_s and σ_a , respectively, which describe the decrease of light intensity per distance and are therefore given in units of inverse meter (or here inverse megameter $1 \text{ Mm}^{-1} = 10^{-6} \text{ m}^{-1}$). For aerosols in the diameter range between 10 nm and $\sim 2.5 \mu\text{m}$, as considered here, both σ_s and σ_a are complex functions of particle size and shape as well as the degree and kind of mixing with other particles (internally and externally mixed) (Bohren and Huffman, 1983; Fuller et al., 1999). While reliable in-situ measurement techniques for light scattering by aerosols have been available for several decades (Heintzenberg and Charlson, 1996), light absorption is by nature a more elusive property, since during the absorption process photons are converted into thermal energy, which makes it impossible to detect them directly. Historically, two main approaches have been applied to experimentally determine σ_a , namely the filter-based attenuation and the difference method (Horvath, 1993). The former involves deposition of aerosols onto a filter substrate and measuring the change in light attenuation through the filter as sample aerosol is deposited. Commercially available instruments of this type are the AethalometerTM (Hansen et al., 1984) and Particle Soot Absorption Photometer (PSAP) (Bond et al., 1999). However, due to aerosol-filter interactions these instruments require site-specific calibration factors (Liousse et al., 1993; Petzold et al., 1997; Ballach et al., 2001; Arnott et al., 2005) that have most frequently been determined by intercomparison measurements with the difference method. The difference method determines light absorption of particles in their suspended state from the difference of extinction and scattering typically measured by an optical extinction cell and an integrating nephelometer, respectively. (Reid et al., 1998; Bond et al., 1999; Weingartner et al., 2003; Arnott et al., 2005). Employing the difference method for field studies is problematic due to the low absorption coefficients encountered under ambient conditions (frequently less than the detection limit of about 20 Mm^{-1}) and the large measurement uncertainties ($>25\%$; Schnaiter et al., 2005) for single

scattering albedos typical for ambient aerosols (larger than ~ 0.7). Recently, Petzold and Schoenlinner (2004) have introduced a novel filter-based absorption technique, the multi-angle absorption photometer (MAAP), that measures not just light transmission through (as the Aethalometer and PSAP) but also angular reflection from an aerosol-laden filter and combines it with a two-stream-approximation radiative transfer scheme to account for filter-particle interactions (Petzold et al., 2005).

All methods described above are not only sensitive to light absorption but also to scattering effects that have to be adequately accounted for to yield reliable absorption values. In contrast, the photoacoustic spectrometer (PAS) (Truex and Anderson, 1979) is sensitive to absorption only, since it measures an acoustic signal that originates from the thermal response of irradiated particles due to light absorption. The measurements are performed on aerosols in their suspended state, i.e., filter artifacts do not occur. Recent laboratory calibration experiments with kerosene- and spark-generated soot have shown excellent agreement (better than 10%) between the PAS and the difference method (Schnaiter et al., 2005; Sheridan et al., 2005; Virkkula et al., 2005). In addition, unlike any other absorption technique the PAS can be calibrated on-site with a calibration gas (Arnott et al., 2000).

This study is the first of two parts on spectral light absorption by ambient aerosols in the Amazon Basin measured during the SMOCC field campaign from 9 September to 14 November 2002. Part I reports on the field intercomparison of a 7- λ Aethalometer ($\lambda=450$ to 950 nm) and a 1- λ PSAP (565 nm) with a 1- λ photoacoustic spectrometer (PAS, 532 nm) as reference device. The principle, operation and performance of all three absorption instruments are briefly discussed and the multiple scattering and filter loading correction for the Aethalometer and PSAP are determined. Finally, for the latter two devices, the effects of relative humidity, single scattering albedo and gaseous adsorption onto the filter substrate are investigated. A detailed discussion of the spectral absorption properties of Amazonian aerosol will be provided in the Part II of this paper.

2 Experimental

2.1 Measurement site and period

From 9 September to 14 November 2002 the Large Scale Biosphere-Atmosphere Experiment in Amazonia – Smoke, Aerosols, Clouds, Rainfall and Climate (LBA-SMOCC) campaign was conducted in the state of Rondônia, Brazil (Andreae et al., 2004). The measurement station was located on the Fazenda Nossa Senhora Aparecida (10.76° S, 62.32° W, 315 m a.s.l.), a pasture site in the south-western part of the Amazon Basin about 50 km north-west of Ji-Parana (10.88° S, 61.85° W, 235 m a.s.l.; $\sim 110\,000$ inhabitants) (Andreae et al., 2002). While the area around

FNS is predominantly grassland, the site is affected by the widespread vegetation fires due to fire-assisted land-clearing activities in the Amazon Basin during the dry season (June–October). The measurement period was selected such that both dry season and wet season data could be collected. Here we will distinguish between three periods: the dry period from 9 September to 8 October (end of dry season), a dry-to-wet transition period from 9 October to 30 October, and the wet period from 1 November to 14 November (beginning of wet season). While the dry period is heavily influenced by biomass burning events, this burning signature is significantly reduced in the transition period and reaches even lower levels in the wet season.

2.2 Setup

A comprehensive suite of aerosol, gas phase and meteorological parameters was measured during the SMOCC campaign. Here we focus on instrumentation for aerosol light absorption measurements. As mentioned above, aerosol light absorption was measured with three different instruments: a 1- λ photoacoustic spectrometer (PAS, 532 nm), a 1- λ Particle Soot Absorption Photometer (PSAP, Radiance Research, 565 nm) and a 7- λ Aethalometer (AE30, Magee Scientific, 450 to 950 nm). In addition, two integrating 1- λ nephelometers (Radiance Research, M903, 545 nm) were used to measure aerosol light scattering (Chand et al., 2006).

The aerosol inlets (Rupprecht & Patashnick; inlet for the TEOM 1400) were located ~ 1 m above the roof top of the instrument hut (~ 7 m above the ground). They were equipped with a 1.5 or 10 μm impactor, i.e., we sampled particulate matter either below 1.5 or 10 μm aerodynamic diameter (PM1.5 and PM10, respectively).

The PSAP, PAS and one of the nephelometers were sampling from the same Rupprecht & Patashnick PM10 inlet equipped with an additional 1.5 μm impactor. Prior to particle detection the aerosol was dried to $RH < 45\%$ by a Nafion membrane counter-flow drier (Permapure, Inc.) and then passed through the 1.5 μm impactor. The particle loss in the Nafion drier ($< 5\%$ for $50 \text{ nm} < D_p < 700 \text{ nm}$) and the cut-off characteristics of the 1.5 μm impactor were experimentally determined after the campaign with dry ammonium sulfate particles. Both the absorption and scattering coefficient (σ_a and σ_s , respectively) were corrected for line losses (on average $\sim 2.5\%$) utilizing the measured particle loss, the dry particle size distributions and the Mie code described by (Guyon et al., 2003a). Particle loss in the connecting stainless steel transport lines was considered negligible in the size range of interest for aerosol optical properties (30 nm to 10 μm diameter), since for each instrument the length of the connecting tubing was below 10 m and the flow conditions were kept laminar. All flow rates were regularly calibrated to an estimated accuracy of about 2% with a positive displacement flow meter. The sampling flow rates of the PSAP, PAS and nephelometer were 0.2–0.4, 0.8 and 1.0–1.2 L min^{-1} , respec-

tively. Although the flow rates for the PSAP and nephelometer were smaller than specified by the manufacturer (to allow for more efficient drying of the sample flow and longer lifetime of the PSAP filter) we have not seen a systematic change in instrument response when the flow rate was increased to manufacturer specifications (PSAP: 1 to 4 L min^{-1} ; nephelometer: 10 L min^{-1}). The time resolution of the PSAP, Aethalometer, PAS and nephelometers was 1 min, ~ 15 min, 10 s, and 1 min, respectively. For the calibration of the PSAP and Aethalometer using the PAS, we converted all data to the sampling rate of the slowest device, the Aethalometer. The Aethalometer and the other nephelometer were operated with non-dried PM10 aerosol at a flow rate of about 6.6 L min^{-1} and 7 L min^{-1} , respectively. We will refer to these operating conditions as “ambient”, although the term “non-dried” is more accurate, since the absence of an active drying procedure led to operational relative humidities (RH) that were somewhat lower than ambient RH due to slightly elevated instrument temperatures, especially during nighttime. While during the dry and warm daytime conditions ambient and instrument RH were within a few percent, the differences reached about 20% during nighttime when ambient RH was close to 100%, but instrument RH only reached about 80%.

Since the Aethalometer and the PAS were operated under different RH conditions and from inlets with different cut-off diameters (PAS: PM1.5; Aethalometer: PM10), we have to consider these differences when comparing the Aethalometer with the PAS. The effect of RH on Aethalometer performance is negligible as will be discussed below. Regarding the size cut-off we utilized the size-segregated aerosol mass information provided by a collocated MOUDI impactor (Marple et al., 1991). The relative mass contribution of the 1.8 to 10 μm size segment (stage 2+3 of the MOUDI impactor) to PM10 (stages 2 to 10) was on average 7.6% (dry period) and 14.9% (transition period). Considering that most of the absorbing material (black carbon) is found in PM1.5 and that the mass specific absorption cross section decreases with size for supermicron particles (Horvath, 1993), the cut-off-related systematic difference between Aethalometer and PAS signal is expected to be considerably less than 8 and 15% for the dry and transition period, respectively. We will see below that these difference are negligible compared to other effects. Since the PSAP and PAS were operated from the same inlet, no such considerations are necessary for the PSAP. Unless stated otherwise, all data are referenced to 1000 hPa and 298.2 K.

2.3 Photoacoustic spectrometer (PAS)

2.3.1 Principle of operation

The photoacoustic spectrometer (PAS) determines aerosol light absorption by converting the absorbed energy into an acoustic wave detected by a sensitive microphone (Terhune and Anderson, 1977). While passing aerosol through

an acoustic resonator, a power-modulated laser periodically heats the aerosol, which leads to periodic thermal expansions and hence pressure pulses (acoustic wave). Using a calibrated microphone the pressure amplitude P_m of this acoustic wave is measured and the nominal absorption coefficient $\sigma_{\text{PAS,raw}}$ can be calculated according to (Rosencwaig, 1980)

$$\sigma_{\text{PAS,raw}} = P_m \frac{\pi^2 A_{\text{res}} f_0}{P_L \tilde{Q} (\gamma - 1)}, \quad (1)$$

where A_{res} , f_0 , and \tilde{Q} are the cross sectional area, the acoustic resonance frequency, and the quality factor of the resonator, respectively, and P_L and γ are the modulated average laser power and the ratio of the isobaric and isochoric specific heats of the carrier gas ($\gamma_{\text{air}}=1.4$), respectively.

2.3.2 Technical details

The device used here, a refined version of the PAS described by Arnott et al. (1999), was optimized for atmospheric applications by maximizing the signal to noise ratio. The PAS utilizes a frequency-doubled diode-laser-pumped Nd:YAG laser ($\lambda=532$ nm), which is power-modulated by a chopper at the resonance frequency of the acoustic resonator ($f_0=1500$ Hz). The modulated laser power of $P_L \cong 60$ mW is continuously monitored (after passing through the resonator) by a photodiode mounted on an integrating sphere. The length and cross-sectional area of the resonator are 24.86 cm and 2.18 cm², respectively. To avoid potential systematic errors due to temperature and pressure drifts in f_0 and \tilde{Q} (~ 75), both f_0 and \tilde{Q} are continuously measured and optimized for acoustic resonance utilizing a piezoelectric disc. Since P_m , P_L , f_0 and \tilde{Q} are directly measured by the PAS, all parameters of Eq. (1) are known and the absorption coefficient can be calculated without any device-specific calibration factor as typical for most filter-based absorption techniques. From the experimental uncertainties in these measurement parameters we estimated the overall uncertainty of $\sigma_{\text{PAS,raw}}$ as 5%. To optimize the signal to noise ratio (and hence the lower detection limit) the acoustic noise was minimized passively by (1) using absorbing materials, (2) avoiding turbulent flow conditions and sharp bends in the connecting tubing, (3) installing an acoustic filter at the inlet of the resonator (two volumes with different acoustic resonance frequency, i.e., low and high pass filters in series) and (4) acoustically isolating the sample pump from the resonator by a critical orifice. It is also noteworthy that, while the sample flow rate (here 0.8 L min⁻¹) affects the response time of the PAS (here: <10 s), it does not enter Eq. (1), i.e., the sample flow rate is irrelevant for the measured σ_a .

2.3.3 Calibration and intercomparison with difference method

In general the acoustic signal of the PAS may originate from absorbing particulate or gaseous components. Hence, in

contrast to filter-based absorption techniques, photoacoustic sensors can be calibrated utilizing the well-known absorption properties of gaseous components. Recently, Arnott et al. (2000) have introduced a calibration procedure, which does not require any information beyond the data stream provided by the PAS itself. To rationalize this method it is important to note that during normal operation of the PAS, the extinction of the laser light in the acoustic resonator is due to both absorption and scattering effects by particulate and gaseous components, while the acoustic signal responds to absorption only. By introducing a particle-free absorbing calibration gas (here ~ 1000 ppm NO₂ in air; $\sigma_a \sim 330\,000$ Mm⁻¹) particulate effects are eliminated and the extinction of the laser light and the acoustic signal depend on gaseous effects only. Since for the calibration gas, scattering is small (Rayleigh scattering coefficient ~ 10 Mm⁻¹ at normal conditions) compared to σ_a , both the mitigation of the laser light and the acoustic signal solely depend on absorption, i.e., Lambert-Beer's law can be used to derive a reference absorption value σ_{LB}

$$P_L = P_{L,0} \exp(-\sigma_{LB} L), \quad (2)$$

where L ($=0.2486$ m) is the optical length of the resonator and P_L and $P_{L,0}$ are the laser intensities with and without NO₂ in the resonator, respectively, that can be determined by the photomultiplier of the PAS. It is evident from Eq. (2) that σ_{LB} is completely independent of the photoacoustic signal of the PAS and that neither the concentration nor the absorption cross section of NO₂ is required. The only requirement is that, on the one hand, the NO₂ concentration is large enough to neglect Rayleigh scattering and to introduce a measurable change in P_L and, on the other hand, the NO₂ concentration is small enough not to exceed the linear response range of the microphone. As mentioned above, for the SMOCC campaign, we used ~ 1000 ppm of NO₂ in synthetic air as calibration gas, which corresponded to an absorption coefficient of $\sigma_{\text{PAS,raw}} \sim 330\,000$ Mm⁻¹. By progressively diluting the calibration gas with filtered air, we confirmed that the microphone was linear up to at least 330 000 Mm⁻¹, i.e., over a dynamic range of more than five orders of magnitude.

Figure 1 illustrates a PAS calibration cycle where each of the data points represents an averaging period of ~ 6 s. The zero signal is determined with particle free air. When the particle free air is replaced by the calibration gas (at measurement point 10), the photoacoustically determined absorption coefficient ($\sigma_{\text{PAS,raw}}$; see Eq. 1) increases abruptly from 0.5 ± 1.2 Mm⁻¹ to $330\,000 \pm 3000$ Mm⁻¹ (average and standard deviation), while the laser intensity (after passing through the resonator) decreases from $P_{L,0} = 61.631 \pm 0.009$ mW to $P_L = 56.870 \pm 0.014$ mW, which according to Eq. (2) corresponds to $\sigma_{LB} = 323\,000 \pm 1000$ Mm⁻¹. When at measurement point 24 the PAS is purged with particle free air again, both $\sigma_{\text{PAS,raw}}$ and P_L return to their initial values. Comparing $\sigma_{\text{PAS,raw}}$ and σ_{LB} we find that $\sigma_{\text{PAS,raw}}$ is 2.2% larger

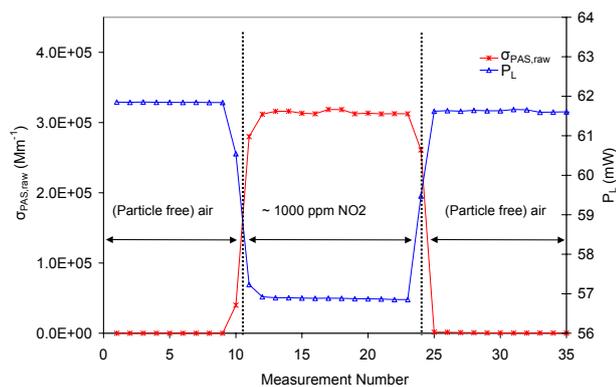


Fig. 1. Response of the PAS during a NO_2 calibration cycle, where $\sigma_{\text{PAS,raw}}$ (stars) and P_L (triangles) are the photoacoustically determined absorption coefficient and the laser power, respectively.

than σ_{LB} (at a precision of 1.0%), which is well within the estimated overall uncertainty of the PAS (5%). Again we note that this simple two-point calibration procedure does not rely on any external calibration standard nor does it require exact knowledge of the NO_2 concentration or any other information not provided by the data stream of the PAS.

As an additional measure of quality assurance we performed laboratory experiments with various types of aerosols comparing the absorption coefficient of the PAS to the difference of extinction (σ_e) and scattering (σ_s) determined by an optical extinction cell, the Long Path Optical Extinction Spectrometer (LOPES), and an integrating nephelometer (TSI, model 3563), respectively. To optimize the accuracy of the difference method, systematic biases due to e.g. the finite acceptance angle of the extinction cell and the angular non-idealities of the nephelometer were taken into account (Schnaiter et al. 2005). Figure 2 depicts the measured absorption coefficients for pure soot particles (solid symbols) and soot particles coated with non-absorbing materials (organic and inorganic; open symbols), where the organic coating was produced by ozonolysis of α -pinene, which among other organic compounds generates pinic and pinonic acids (Saathoff et al., 2003). The absorption coefficients measured by the PAS and the difference method agree well for both pure soot particles (Diesel and spark-generated [PALAS] soot) and coated soot particles (slope = 0.972 ± 0.022). This confirms the results from a previous laboratory study which was performed on pure soot and biomass burning aerosols (Schnaiter et al., 2005).

2.3.4 Data reduction and accuracy

The main sources for systematic uncertainties of the PAS under field conditions are zero point instabilities and the cross-sensitivity to ambient NO_2 . Since this may result in a variable zero-point offset, the instrument offset was repeatedly determined by zero calibrations using filtered (particle-free)

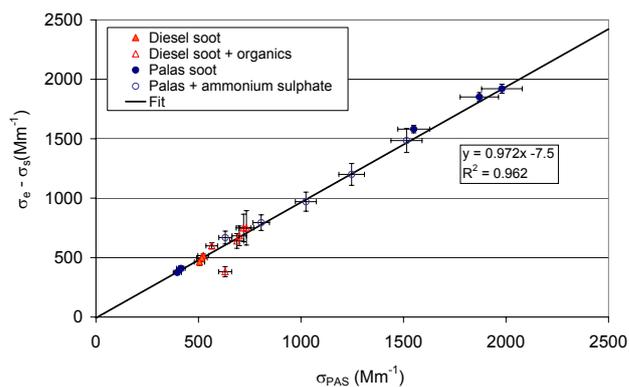


Fig. 2. Comparison of absorption coefficients determined by the PAS and the difference method (extinction minus scattering, $\sigma_e - \sigma_s$) for both pure (Diesel and spark-generated (PALAS) soot) and coated soot particles (internally mixed aerosols).

ambient air for 10 to 30 min (at least) twice a day. We corrected for the NO_2 cross-sensitivity utilizing the ambient NO_2 mixing ratios continuously measured by a Model 42CTL NO/NO_x monitor (Thermo Environment Instruments Inc.) (Kirkman et al., 2002). For the same Nd:YAG laser as used here, Arnott et al. (2000) determined a NO_2 specific absorption coefficient of $0.306 \pm 0.015 \text{ Mm}^{-1} \text{ ppb}^{-1}$ ($156\,000 \pm 1000 \text{ Mm}^{-1}$ for $509\,000 \pm 25\,000 \text{ ppb}$ of NO_2) at 846 hPa and 294.7 K. Hence, the NO_2 -induced PAS offset can be expressed as

$$\begin{aligned} \sigma_{\text{NO}_2} &= 0.306 \frac{p}{846 \text{ hPa}} \frac{294.7 \text{ K}}{T} c_{\text{NO}_2} \text{ Mm}^{-1} \text{ ppb}^{-1} \\ &= B_{\text{NO}_2} \frac{p}{T} c_{\text{NO}_2}, \end{aligned} \quad (3)$$

where the lump constant B_{NO_2} equals $0.107 \pm 0.005 \text{ K hPa}^{-1} \text{ ppb}^{-1} \text{ Mm}^{-1}$ and p , T and c_{NO_2} are the operating pressure, temperature and NO_2 (volume) mixing ratio, respectively. Based on these considerations each time layer of the PAS data was corrected for zero offset and NO_2 sensitivity according to

$$\sigma_{\text{PAS}} = \sigma_{\text{PAS,raw}} - \sigma_0 - B_{\text{NO}_2} \left(\frac{p}{T} c_{\text{NO}_2} - \frac{p_0}{T_0} c_{\text{NO}_2,0} \right), \quad (4)$$

where $\sigma_{\text{PAS,raw}}$ is given by Eq. (1) and σ_0 and $c_{\text{NO}_2,0}$ are $\sigma_{\text{PAS,raw}}$ and c_{NO_2} during the PAS zero calibration, respectively. Since the zero calibration is performed with particle-free, but not NO_2 denuded air, it is also necessary to include the $c_{\text{NO}_2,0}$ term which accounts for the NO_2 bias in σ_0 . During the dry period of the SMOCC campaign, the period with the largest NO_2 contribution, an average of 6.5% of the PAS signal could be attributed to NO_2 . However, since not the absolute NO_2 concentration but the deviation from $c_{\text{NO}_2,0}$ enters Eq. (4), the NO_2 correction term was typically less than 1% of σ_{PAS} , except for a few instances where sudden drastic changes in pollution levels intermittently enhanced the NO_2 correction term to up to 20%.

In addition to NO₂ interference, the PAS data may be biased by (partial) aerosol volatilization due to laser-induced particle heating, since the latent heat of vaporization would reduce the amount of laser energy generating the acoustic wave and hence reduce the apparent σ_{PAS} (Raspet et al., 2001). For a PAS similar to the one used here, Arnott et al. (2003) showed that for atmospheric aerosol with a deliquescence point of $RH \sim 60\%$ the volatilisation effect was negligible ($< 10\%$) up to RH levels of about 80%. Considering the relatively small average hygroscopic diameter growth during SMOCC (< 1.04 for $RH < 45\%$; (Rissler et al., 2006)), we anticipate no bias in the PAS signal due to water evaporation. This is corroborated by the absence of a phase shift between PAS microphone signal and oscillating laser power during the SMOCC campaign, which also indicates a negligible PAS bias due to mass transfer effects (Arnott et al., 2003).

Based on these considerations we estimate the accuracy of the PAS under field conditions as better than 10% (95% confidence level) for $\sigma_{\text{PAS}} > 10 \text{ Mm}^{-1}$, which is larger than the 5% accuracy achieved under controlled laboratory conditions, since it includes the uncertainties due to unavoidable instabilities in operating conditions. For averaging periods of 5, 15 and 60 min, the instrument noise (precision) was 1.1, 0.7 and 0.4 Mm^{-1} (95% confidence level), respectively, which results in a lower detection limit (three times the 1σ noise level) of 1.6, 1.1 and 0.6 Mm^{-1} , respectively.

2.4 Aethalometer

2.4.1 Principle

The 7- λ Aethalometer (AE30, Magee Scientific) measures light attenuation ATN at seven wavelengths (450, 571, 590, 615, 660, 880, and 950 nm, where the 571 nm channel had to be discarded for reasons discussed below) through an aerosol-laden quartz filter based on (Hansen et al., 1984)

$$ATN = 100 \ln \left(\frac{I_0}{I} \right), \quad (5)$$

where I and I_0 are the light intensities transmitted through the particle-laden and a blank spot of the filter, respectively. If aerosol is deposited onto the filter for a time period Δt , the attenuation coefficient σ_{ATN} is given by

$$\sigma_{\text{ATN}} = \frac{A}{100Q} \frac{\Delta ATN}{\Delta t}, \quad (6)$$

where A is the area of the aerosol-laden filter spot and Q is the volumetric sampling flow rate. The standard output protocol of the manufacturer provides equivalent black carbon mass concentration BC_{ATN} (g m^{-3}), which is determined from σ_{ATN} according to

$$BC_{\text{ATN}} = \frac{\sigma_{\text{ATN}}}{\alpha_{\text{ATN}}} \quad (7)$$

and

$$\alpha_{\text{ATN}} \left[\text{m}^2 \text{g}^{-1} \right] = 14\,625 / \lambda \text{ [nm]}, \quad (8)$$

where the spectral mass specific attenuation cross-section α_{ATN} is based on a calibration at 880 nm utilizing the Malissa-Novakov method, a solvent-based thermal desorption method for elemental carbon analysis (Gundel et al., 1984). Since the reliability of thermal desorption methods is still under debate (Schmid et al., 2001), we avoid the resulting uncertainties by limiting our investigation to σ_{ATN} , the primary measurement parameter of the Aethalometer. Hence, each 15 min BC_{ATN} value was converted into σ_{ATN} according to Eqs. (7) and (8).

2.4.2 Relating attenuation and absorption

It is well-known that σ_{ATN} is generally larger than σ_a due to optical interactions of the filter substrate with the deposited aerosol (Petzold et al., 1997; Kopp et al., 1999; Ballach et al., 2001; Weingartner et al., 2003; Arnott et al., 2005). The most significant filter-particle interactions and the resulting biases are: (1) multiple scattering of light at the filter fibers enhances the optical path length and hence imposes a positive bias on σ_{ATN} , (2) enhanced absorption of scattered light with increasing filter loading reduces the optical path length and hence reduces σ_{ATN} , and (3) the filter reflectance (scattering in backwards hemisphere) and hence the measured ATN depends on the optical properties of the deposited particles (bias in σ_{ATN} depends on physico-chemical properties of the particles).

Recently, Weingartner et al. (2003) (henceforth referred to as W2003) have shown that the absorption coefficient determined by the Aethalometer (σ_{aeth}) can be expressed as

$$\sigma_{\text{aeth}} = \frac{\sigma_{\text{ATN}}}{C R(ATN)}, \quad (9)$$

where the constant factor C (≥ 1) corrects for multiple light scattering effects within the filter and $R(ATN)$ (≤ 1) accounts for the “shadowing” effect due to filter loading (decrease in Aethalometer sensitivity). Since the shadowing factor (R) is small for lightly loaded filters ($ATN < 10$), C can be determined from (W2003)

$$C = \frac{\sigma_{10}}{\sigma_{\text{PAS}}}, \quad (10)$$

where σ_{10} represents all σ_{ATN} values with $ATN < 10$ (i.e., $R \approx 1$) and σ_{PAS} is the PAS-based (reference) absorption coefficient. Furthermore, the loading correction can be expressed as (W2003)

$$R(ATN) = \left(\frac{1}{f} - 1 \right) \frac{\ln ATN - \ln 10}{\ln 50 - \ln 10} + 1, \quad (11)$$

where the shadowing factor f is a parameter that depends on the type of aerosol and ATN is measured directly by the Aethalometer (W2003). Setting $ATN=10$ for all ATN values

smaller than 10, R is equal to unity, i.e. the loading is so small that there is no effect of filter loading on the performance of the Aethalometer. On the other hand, for $\text{ATN} > 10$, R decreases with ATN . The steepness of this decrease depends on the factor f . As seen from Eq. (11), if $f=1$ we find that R is equal to unity independent of ATN , i.e. the aerosol deposited onto the filter has no effect on the Aethalometer performance. On the other hand, if $f > 1$ we find that R becomes smaller than unity, since the absorbing components of the deposited aerosol reduce the amount of multiple scattering within the filter matrix, i.e., f can be described as “shadowing” parameter. For the SMOCC data, the factor f can be determined by fitting Eq. (11) to the measured R values given by

$$R_{\text{meas}}(\text{ATN}) = \frac{\sigma_{\text{ATN}}(\text{ATN})}{\sigma_{\text{PAS}}C}. \quad (12)$$

Here R_{meas} can be interpreted as the loading dependent Aethalometer sensitivity that is unity for a pristine filter and decreases with increasing filter loading.

2.4.3 Effect of aerosol scattering on attenuation

W2003 found no significant dependence of σ_{aeth} (<1%) on the scattering component of the sample aerosol (σ_s) for ammonium sulfate. The enhanced scattering effect (up to 5%) for organic carbon particles produced by ozonolysis of α -pinene was interpreted as an artifact due to a small (but unknown) absorption component of the organic carbon particles. On the other hand, Arnott et al. (2005) (henceforth referred to as A2005) reported a dependence of the Aethalometer signal on σ_s . The reason for this discrepancy is unknown, but we will see below, that it is irrelevant for the calibration performed here. Based on their findings A2005 suggested the following expression for the Aethalometer derived absorption coefficient σ_{aeth}

$$\sigma_{\text{aeth}} = \frac{\sigma_{\text{ATN}} - m_s \sigma_s}{C^* R(\text{ATN})}, \quad (13)$$

where m_s represents the fraction of the aerosol scattering coefficient σ_s that is erroneously interpreted as absorption and C^* and $R(\text{ATN})$ are the multiple scattering and loading correction, respectively. The asterisk on C^* indicates that the magnitude of the multiple scattering correction introduced by A2005 differs from that defined by W2003 (C in Eq. 9) due to the scattering term. By equating Eqs. (13) and (9) (both equations refer to the absorption coefficient derived from the Aethalometer data) the relationship between C and C^* can be expressed as

$$C^* = \frac{C(\sigma_{\text{ATN}} - m_s \sigma_s)}{\sigma_{\text{ATN}}}. \quad (14)$$

Eq. (14) indicates that C^* represents the multiple scattering correction C corrected for aerosol scattering. Using the definition of the single scattering albedo

$$\omega_0 = \frac{\sigma_s}{\sigma_s + \sigma_{\text{aeth}}}, \quad (15)$$

we can substitute σ_s in Eq. (14) by

$$\sigma_s = \frac{\omega_0}{1 - \omega_0} \sigma_{\text{aeth}}, \quad (16)$$

yielding

$$C^* \approx C \left[1 - \frac{m_s \omega_0}{C(1 - \omega_0)} \right], \quad (17)$$

where the approximation $\sigma_{\text{ATN}}/\sigma_{\text{aeth}} \approx C$ was used, i.e., we neglected the loading factor R (see Eq. 9), which is close to unity (0.9 ± 0.1 at 532 nm) as will be shown below. Equation (17) shows that the aerosol scattering effect (m_s term) increases with ω_0 and decreases with C (multiple scattering from the filter matrix). Obviously, in absence of aerosol scattering effects ($m_s=0$ and/or $\omega_0=0$), $C^*=C$.

Provided m_s is known, Eq. (13) instead of Eq. (9) could be used for the field calibration of an Aethalometer. As mentioned above, while W2003 found no significant scattering effect ($m_s < 0.01$), A2005 reported much higher m_s values of 0.052 (at $\lambda=521$ nm) for ammonium sulfate as challenge aerosol. Due to the considerable uncertainty in m_s , we will base our calibration efforts on Eq. (9), but use Eq. (17) to estimate the effect of aerosol scattering on the multiple scattering (filter matrix) correction C .

2.4.4 Spectral dependence of calibration factors

Since none of the seven Aethalometer channels ($\sigma_{\text{ATN},i}$ with $i=1, 2, \dots, 7$) matches the wavelength of the PAS ($\lambda_{\text{PAS}}=532$ nm), σ_{aeth} (and hence σ_{ATN}) has to be converted to λ_{PAS} according to

$$\sigma_{\text{ATN}} = \sigma_{\text{ATN},0} \left(\frac{\lambda_{\text{PAS}}}{\lambda_0} \right)^{-\alpha_{\text{ATN}}}, \quad (18)$$

where the attenuation Ångström exponent α_{ATN} was calculated from two Aethalometer channels using

$$\alpha_{\text{ATN}} = - \frac{\log \sigma_{\text{ATN},0} - \log \sigma_{\text{ATN},1}}{\log \lambda_0 - \log \lambda_1}. \quad (19)$$

Analogous to α_{ATN} , the Ångström exponents of absorption (α_a) and scattering (α_s) are defined by replacing σ_{ATN} by σ_a and σ_s in Eq. (19), respectively. Here, λ_0 and λ_1 were equal to 590 and 450 nm, respectively, unless stated otherwise. For illustration of the λ -dependence of σ_{ATN} and ATN , Fig. 3 depicts a time series of σ_{ATN} (lines) and ATN (triangles) for three of the seven Aethalometer wavelengths, namely 450, 590 and 880 nm, represented by the colors blue, green and red, respectively. Once ATN (590 nm) reaches ~ 75 , the Aethalometer automatically forwards the filter tape, ATN is set to zero (here at about 04:00) and the new filter spot remains exposed to the sample flow until ATN (590 nm) reaches ~ 75 again. Both σ_{ATN} and ATN increase with decreasing wavelength. The average attenuation Ångström exponent α_{ATN} , derived from Eq. (19), was about 1.5. Please

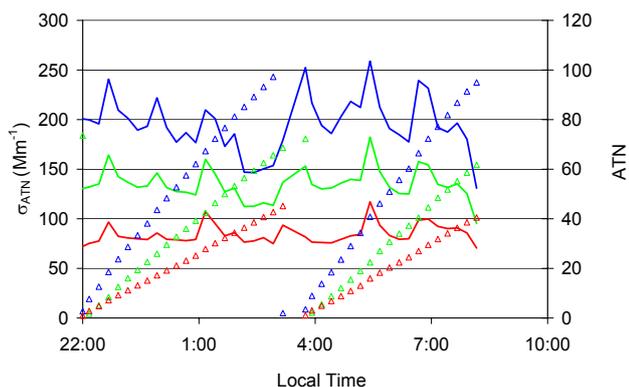


Fig. 3. Time series of attenuation coefficients (lines) and attenuation (triangles, secondary axis) for $\lambda=450, 590$ and 880 nm (blue, green, and red, respectively).

note that the sensitivity of the Aethalometer decreases with increasing ATN. This can be seen from Fig. 3, where at about 04:00 the filter tape forwards and the σ_{ATN} traces show a step-increase that is most pronounced for $\lambda=450$ nm, the trace with the largest ATN values. We will correct for this loading effect as described below.

Since our reference instrument operated at $\lambda_{\text{PAS}}=532$ nm, we can only derive the Aethalometer correction factors f and C for λ_{PAS} . Application of these factors to all seven Aethalometer channels may introduce systematic biases especially for $\lambda=950$ nm, the wavelength most different from 532 nm. Hence, it is important to investigate the spectral dependence of f and C .

For a wide variety of soot particles (internally/externally mixed; fresh/aged), W2003 showed that the dependence of the filter loading correction R on ATN is independent of λ , i.e., the shadowing factor f is a weak function of wavelength (see Eq. 11). While the mean f values ranged from 1.15 to 1.65 depending on the type of challenge aerosol, f was constant to within 0.09 (absolute) for a given aerosol type (W2003; excluding the atmospherically not relevant case for PALAS soot), i.e., for a given ATN value, R was constant to within 11.8%. Obviously, for $\text{ATN}=0$, R is equal to unity independent of f and hence, averaged over an entire filter cycle, the effect of the wavelength dependence of f on R is considerably smaller than 10% even for the most affected channel (450 nm). Thus using a λ -independent f value introduces Aethalometer biases much smaller than the calibration uncertainty ($\sim 20\%$; see Sect. 3.2.5) in any of the AE30 channels and hence, the wavelength dependence of f is negligible for most atmospheric applications. In response to some recent misinterpretations of this finding (Kirchstetter et al., 2004) we emphasize again that this does not mean that the loading correction (R) itself is wavelength independent. As seen from Eq. (11) R increases with ATN and since ATN increases with decreasing λ as seen from Fig. 3, R increases towards the UV range.

Regarding the multiple scattering correction C , W2003 found only a minor difference of $<10\%$ when comparing the 450 and 660 nm channels and they suggested that C can be considered constant. A similarly modest dependence of C^* on λ (5% increase from 470 to 660 nm) was reported by A2005. However, in contrast to W2003, A2005 reported a non-negligible aerosol scattering correction factor (m_s), i.e., C and C^* are not directly comparable. Hence, one should convert C^* into the overall correction factor C as given by Eq. (17)

$$C = C^* + m_s \frac{\omega_0}{1 - \omega_0}. \quad (20)$$

A2005 provide λ -specific C^* and m_s values for a slightly different set of wavelengths as used here (see their Table 1 and our Table 1; they referred to C^* and m_s as M and α , respectively), but our C^* and m_s values at 532 nm are well approximated by their 521 nm values, i.e., for the purpose of this study we assume $C_{532}^* = C_{521}^*$ and $m_{s,532} = m_{s,521}$. Using Eqs. (15) and (16) we can describe the wavelength dependence of ω_0 as

$$\begin{aligned} \omega_{0,\lambda} &= \frac{\sigma_{s,\text{ref}} \left(\frac{\lambda}{\lambda_{\text{ref}}} \right)^{-\alpha_s}}{\sigma_{s,\text{ref}} \left(\frac{\lambda}{\lambda_{\text{ref}}} \right)^{-\alpha_s} + \sigma_{a,\text{ref}} \left(\frac{\lambda}{\lambda_{\text{ref}}} \right)^{-\alpha_a}} \\ &= \frac{\omega_{0,\text{ref}} \left(\frac{\lambda}{\lambda_{\text{ref}}} \right)^{-\alpha_s}}{\omega_{0,\text{ref}} \left(\frac{\lambda}{\lambda_{\text{ref}}} \right)^{-\alpha_s} + (1 - \omega_{0,\text{ref}}) \left(\frac{\lambda}{\lambda_{\text{ref}}} \right)^{-\alpha_a}}, \end{aligned} \quad (21)$$

where we assumed that σ_s and σ_a scale according to $\lambda^{-\alpha_s}$ and $\lambda^{-\alpha_a}$, respectively, with α_s and α_a being the Ångström exponents for scattering and absorption, respectively. Here, the reference wavelength λ_{ref} is 532 nm (or 521 nm, if the data by A2005 are used). For the SMOCC data, $\omega_{0,\text{ref}}=0.92$ and $\alpha_s=2$ (Chand et al., 2006). Choosing a reasonable range of Ångström exponents for absorption ($\alpha_a=1, 1.5$ or 2) (Kirchstetter et al., 2004) we can now calculate C from Eqs. (20) and (21) for wavelengths between 370 and 950 nm as given in Table 1. Obviously, C increases with λ , but the degree of increase depends on α_a . To parameterize this dependence we have plotted $\ln(C)$ versus $\ln(\lambda/\text{nm})$ for $\alpha_a=1, 1.5$ or 2 (see Fig. 4a) and performed a quadratic fit for each α_a value

$$\ln(C) = A (\ln(\lambda/\text{nm}))^2 + B \ln(\lambda/\text{nm}) + D. \quad (22)$$

Since Eq. (22) can be transformed into

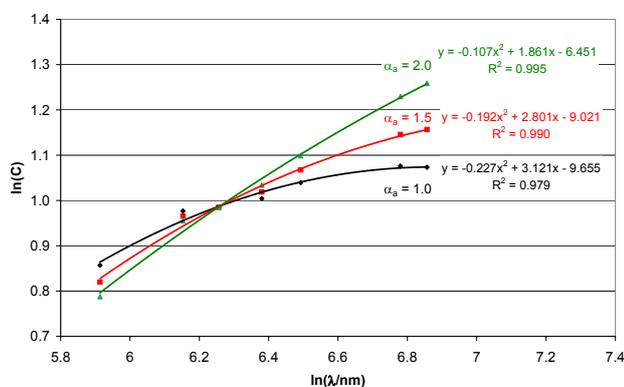
$$\frac{C}{C_{\text{ref}}} = \frac{\lambda^A \ln(\lambda/\text{nm}) + B}{\lambda_{\text{ref}}^A \ln(\lambda_{\text{ref}}/\text{nm}) + B}, \quad (23)$$

the dependence of C on λ can be expressed by the coefficients A and B that depend on α_a , where again for the SMOCC data $\lambda_{\text{ref}}=532$ nm. As seen from Fig. 4b a quadratic

Table 1. Calculation of C according to Eq. (20) using C^* and m_s as given by A2005¹ (Arnott et al., 2005). For ω_0 (521 nm) and α_s we assumed 0.92 and 2, respectively.

α_a	C at various wavelengths (nm)							Ratios of C for various wavelength pairs			
	370	470	521	590	660	880	950	660/470	660/521	521/470	950/521
1	2.355	2.656	2.677	2.730	2.827	2.933	2.925	1.065	1.056	1.008	1.093
1.5	2.270	2.626	2.677	2.770	2.909	3.144	3.179	1.107	1.087	1.019	1.187
2	2.198	2.599	2.677	2.812	3.000	3.420	3.523	1.154	1.121	1.030	1.316

¹ The parameters C^* (and m_s) were taken from the Table 1 of A2005 (they referred to C^* and m_s as M and α , respectively) and given here in ascending order of wavelength (from 370 to 950 nm): 1.813 (0.0335), 2.073 (0.0457), 2.076 (0.0523), 2.104 (0.0616), 2.182 (0.0713), 2.226 (0.1038), 2.199 (0.1148).

**Fig. 4a.** Double logarithmic plot of C versus λ for absorption Ångström exponents $\alpha_a=1.0$ (black), 1.5 (red) and 2.0 (green). The symbols represent the data by A2005 (see Table 1) and the lines are quadratic fits.

fit of A and B versus α_a provides

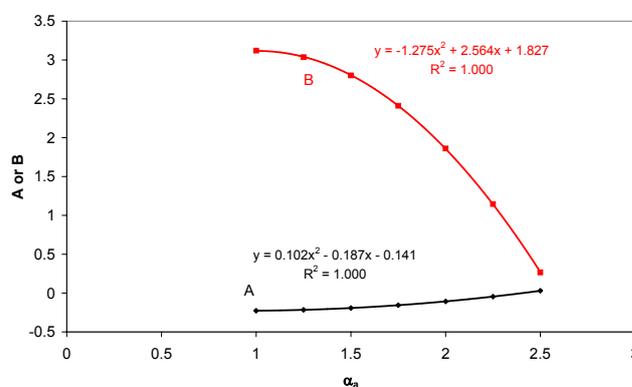
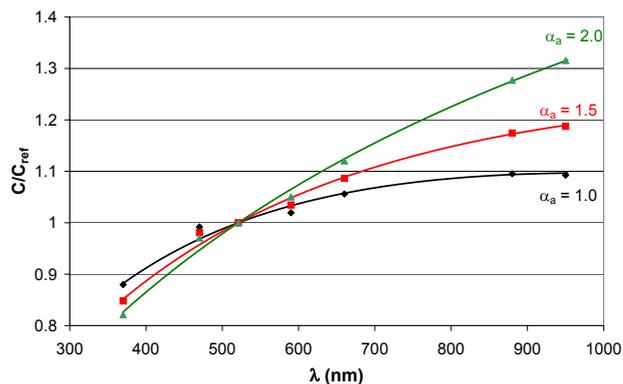
$$A = 0.102 \alpha_a^2 - 0.187 \alpha_a - 0.141$$

$$B = -1.275 \alpha_a^2 + 2.564 \alpha_a + 1.827, \quad (24)$$

respectively. Using Eqs. (23) and (24) we can now determine C for any given λ and α_a . The good agreement between calculated and measured C values is depicted in Fig. 4c that compares the C values listed in Table 1 (normalized to C_{521}) and the corresponding fit curves for $\alpha_a=1, 1.5$ or 2. The maximum deviation between data and fit is less than 3%. We note as caveat that Eq. (24) was derived for the parameters of the SMOCC data ($\omega_{0,\text{ref}}=0.92$ and $\alpha_s=2$), but the procedure described here can be applied to any set of $\omega_{0,\text{ref}}$ and α_s values.

2.5 PSAP

The Particle Soot Absorption Photometer (PSAP; Radiance Research) described by Bond et al. (1999) (henceforth referred to as B1999) measures aerosol light absorption at nominally 565 nm from the light transmitted through an aerosol-laden quartz filter, very similar to the principle of the Aethalometer. Using the difference method (at 550 nm)

**Fig. 4b.** Dependence of the coefficients A and B (see Eq. (23)) on the absorption Ångström exponent (α_a).**Fig. 4c.** Spectral dependence of C normalized to C_{ref} (here C_{521}) for three different absorption Ångström exponents $\alpha_a=1.0, 1.5$ and 2.0. The symbols and lines represent the data (as given in Table 1) and curve fits (see Eqs. (23) and (24)), respectively.

as reference B1999 calibrated the PSAP with pure nigrosin and ammonium sulfate particles as well as internal mixtures of both. Analogous to the calibration equation used for the Aethalometer (see Eq. 13) they found

$$\begin{aligned}\sigma_{\text{PSAP,Bond}} &= \frac{\sigma_{\text{raw,PSAP}} K_Q K_A - K_1 \sigma_s}{K_2} \\ &= \frac{\sigma_{\text{raw,PSAP}} K_Q K_A}{K_2 + K_1 \frac{\omega_0}{1-\omega_0}},\end{aligned}\quad (25)$$

where $\sigma_{\text{raw,PSAP}}$ is the absorption coefficient reported by the PSAP (includes a manufacturer-provided filter loading correction (B1999)), K_Q and K_A are the correction factors for flow rate and sample deposit area, respectively, and the calibration constants K_1 and K_2 are given by $K_1=0.02\pm 0.02$ and $K_2=1.22\pm 0.2$ (95% confidence level), respectively. The last expression in Eq. (25) was derived by applying Eq. (16), where σ_{aeth} was substituted by $\sigma_{\text{raw,PSAP}}$. During SMOCC we used the PSAP output signal ($\sigma_{\text{raw,PSAP}}$) given on a logarithmic voltage scale and the flow rate was artificially set to a constant internal value of 0.5 L min^{-1} , which did not correspond to the true flow rate, but made it simple to correct for the true flow rate Q by using $K_Q=0.5/Q(\pm 3\%)$, where Q is given in L min^{-1} . Similarly the true diameter of the sample deposit spot ($4.86\pm 0.20 \text{ mm}$) deviated from the internally assumed value of 5.1 mm , which resulted in $K_A=(4.86/5.1)^2=0.91\pm 0.07$. Comparing to Virkkula et al. (2005a) (henceforth referred to as V2005a) who measured $K_A=0.97\pm 0.04$ we find that this is within the reported unit-to-unit variability of about 20% (Reid et al., 1998; Anderson et al., 1999; Mader et al., 2002; Wex et al., 2002; Arnott et al., 2003; Guyon et al., 2003a, b). The fact that K_2 does not depend on ATN implies that the loading correction provided by the manufacturer was confirmed at least up to $\text{ATN}=35$ (which corresponds to a transmittance of 0.7). The Bond correction effectively converts the PSAP wavelength from 565 to 550 nm, since their reference device operated at 550 nm. Although B1999 recommend a minimum PSAP filter transmittance of 0.7 ($\text{ATN}=35$), we found no bias down to 0.5 ($\text{ATN}=70$) a result that was also reported by V2005a and Guyon et al. (2003b). Hence we included all data with $\text{ATN}<70$ in the present study. Assuming the uncertainties in K_Q , K_A , K_1 and K_2 are purely random and applying the laws of error propagation to Eq. (25) the accuracy (95% confidence level) of the Bond correction is given by

$$\frac{\Delta\sigma_{\text{PSAP,Bond}}}{\sigma_{\text{PSAP,Bond}}} = \sqrt{\left(\frac{\Delta K_Q}{K_Q}\right)^2 + \left(\frac{\Delta K_A}{K_A}\right)^2 + \frac{(\Delta K_2)^2 + \left(\Delta K_1 \frac{\omega_0}{1-\omega_0}\right)^2}{\left(K_2 + K_1 \frac{\omega_0}{1-\omega_0}\right)^2}}. \quad (26)$$

For an average ω_0 of 0.92 (as applicable for the SMOCC data), we can estimate the accuracy of the Bond corrected SMOCC data as 23% using the uncertainties of K_Q , K_A , K_1 and K_2 given above.

3 Intercomparison and field calibration of PSAP and Aethalometer

For the field calibration of the PSAP and Aethalometer with the PAS, we only included PAS data that showed no statistically significant zero drift for three consecutive zero calibrations, which typically occurred over the course of 24 h. This resulted in about 105 and 95 h of calibration data from the dry (17 September to 8 October) and transition (9 to 30 October) period, respectively. Since we observed no significant dependence of the calibration factors on period, we based the PSAP and Aethalometer calibration on the entire 200 h of PAS data. Due to the low pollution levels throughout the wet period of the SMOCC campaign, the wet period is excluded from the PSAP and Aethalometer calibration, but will be discussed separately below.

3.1 PSAP

As a first approximation we applied the Bond correction (Eq. 25) to the PSAP using the (dry) scattering coefficients (at 545 nm) determined by the nephelometer connected to the same inlet as the PSAP. Performing a linear regression on 5 min averages of $\sigma_{\text{PSAP,Bond}}$ and σ_{PAS} we found a slope of 0.76 ($R^2=0.813$), i.e., changes in PSAP response were on average about 24% lower than changes in σ_{PAS} (data not shown). Accounting for the difference between the reference wavelength of the Bond correction (550 nm) and the PAS (532 nm) using a $\lambda^{-1.5}$ dependence reduced the slope to 0.72, which is outside the 95% confidence level of the Bond correction ($\sim 23\%$). On the other hand, our analysis revealed no systematic dependence of $\sigma_{\text{PSAP,Bond}}$ on either filter loading ($\text{ATN}<70$) or particle single scattering albedo, i.e., the Bond correction adequately accounted for these effects. However, we found a systematic dependence on operating RH and temperature (T) as well as on σ_{PAS} and σ_s . As seen from Fig. 5a, the ratio of $\sigma_{\text{PSAP,Bond}}$ and σ_{PAS} was about constant for $35 < RH < 45\%$ ($\sigma_{\text{PSAP,Bond}}/\sigma_{\text{PAS}}=1.18$; or 1.24, if the PAS is corrected to 550 nm). On the other hand, for low RH between 20% and 30%, $\sigma_{\text{PSAP,Bond}}/\sigma_{\text{PAS}}$ monotonically decreased with RH down to about 0.67 (0.70 for 550 nm). As seen in Fig. 5b, a similar trend is observed for T , where $\sigma_{\text{PSAP,Bond}}/\sigma_{\text{PAS}}$ is positively correlated to T between 24 and 26°C and then remains about constant (slight negative correlation) for $26^\circ\text{C} < T < 31^\circ\text{C}$. It is important to note that during the SMOCC campaign all low RH and T data were gathered during night due to a higher efficiency of the Nafion drier and lower ambient temperatures. Consequently, most of the low and high $\sigma_{\text{PSAP,Bond}}/\sigma_{\text{PAS}}$ values resulted from nighttime and daytime measurements, respectively. Hence, any parameter that shows a significant diel variation will correlate with $\sigma_{\text{PSAP,Bond}}/\sigma_{\text{PAS}}$. This includes σ_{PAS} and σ_s , since we consistently observed elevated pollution levels during nighttime due to the formation of a shallow nocturnal boundary layer (Rissler et al., 2006). Since RH , T , σ_{PAS} and

σ_s are not independently varying parameters, it is not clear which of these parameters (or any other parameter with a pronounced diel variation) is mainly responsible for the observed systematic trend in the PSAP data (see Figs. 5a and b). However, laboratory measurements indicate that $\sigma_{\text{PSAP,Bond}}$ does not depend on σ_{PAS} or σ_s (V2005a). To our knowledge a rigorous investigation of a potential PSAP sensitivity to RH and T has not been performed yet. A more detailed discussion of this issue will be provided below (Sect. 4). For now, we point out that our data do not conclusively identify a specific parameter as cause for the observed bias.

On the other hand, the correlation with both RH and T is strong enough to provide reasonably accurate correction factors for the observed trends. Choosing RH as governing parameter the correction factor K_{RH} was determined by fitting the normalized PSAP data to a second order polynomial (solid line in Fig. 5a)

$$\begin{aligned} K_{RH} &= \sigma_{\text{PSAP,Bond}}/\sigma_{\text{PAS}} \\ &= -0.9212 + 0.1047RH - 0.0013RH^2, \end{aligned} \quad (27)$$

where RH varies between 20 and 43%. For $RH > 43\%$ we used $K_{RH} = K_{43} = 1.18$. Hence, the PSAP-derived absorption coefficient σ_{PSAP} (converted to $\lambda_{\text{PAS}} = 532 \text{ nm}$) was calculated from $\sigma_{\text{PSAP}} = \sigma_{\text{PSAP,Bond}}/K_{RH}$. Figure 5c shows excellent correlation ($R^2 = 0.954$) and agreement (slope = 0.945 ± 0.042) of σ_{PSAP} with σ_{PAS} , where we neglected PAS values smaller than 4 Mm^{-1} to avoid potentially large uncertainties near the lower detection limit. Hence, applying an RH -dependent correction factor (K_{RH}) to the Bond-corrected PSAP data adequately accounts for the PSAP artifacts observed during SMOCC. It is also noteworthy that during nighttime RH (and T) oscillated on a time scale of about 25 min and an amplitude of $\sim 1.0\%$ (absolute) due to fluctuations in the room temperature (air-conditioner turned periodically on and off). These RH (and T) oscillations frequently (not always) induced oscillations in $\sigma_{\text{PSAP,Bond}}$ that were significantly larger than predicted by Eq. (27). We eliminated these oscillations by applying a running average over one oscillation period. The fit parameters given by Eq. (27) are based on these oscillation-corrected data. After removal of these oscillations we did not observe any systematic difference between day and night data that could not be described by the single RH correction equation given above. We estimate the accuracy and precision (95% confidence level) of σ_{PSAP} (532 nm) (5-min averages) as about 15% and 12%, respectively.

3.2 Aethalometer

As mentioned above the operating conditions of the Aethalometer (AE30) differed from those of the PAS in that the Aethalometer was sampling under ambient conditions (no drier) from a $10 \mu\text{m}$ inlet (PAS: $1.5 \mu\text{m}$ impactor). Based

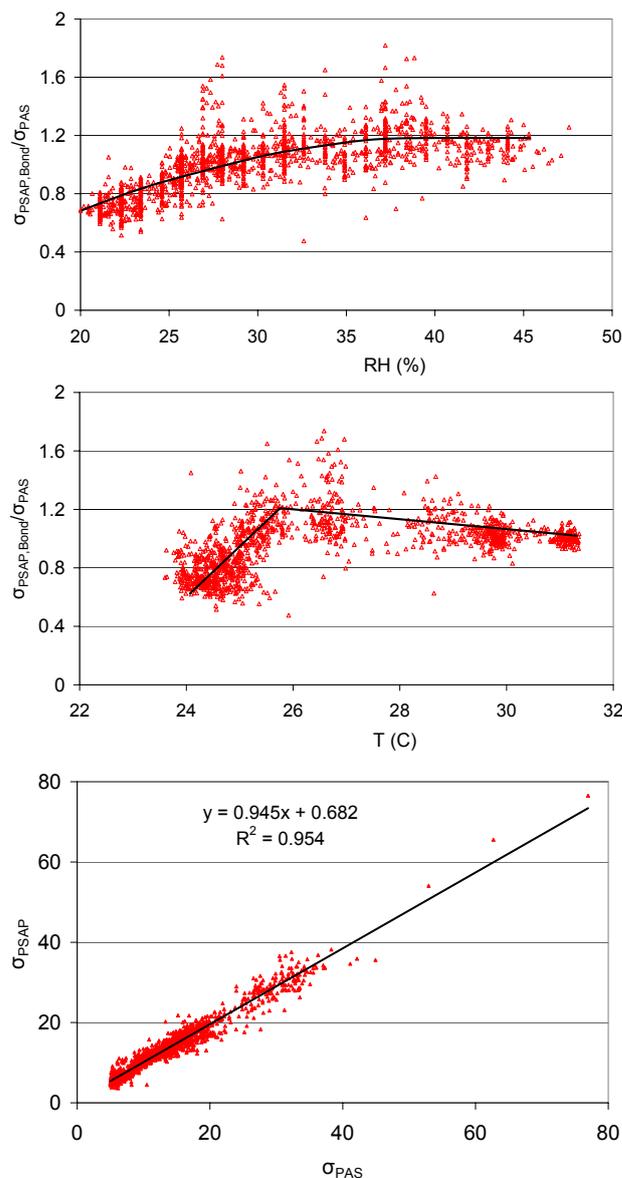


Fig. 5. (a) and (b) Dependence of the Bond corrected normalized PSAP absorption coefficient on relative humidity (RH) and operating temperature (T), respectively. (c) Correlation of the Bond and RH corrected PSAP absorption coefficient σ_{PSAP} with σ_{PAS} .

on MOUDI impactor data the average effect of the different cut-off diameters on total aerosol absorption has been estimated as less than 8 and 15% for the dry and transition period, respectively. A potential systematic effect of RH on the Aethalometer performance will be investigated below.

For the following analysis, the 590 nm channel of the Aethalometer was converted to 532 nm according to Eqs. (18) and (19) using $\lambda_1 = 450$, $\lambda_0 = 590 \text{ nm}$ and $\lambda_{\text{PAS}} = 532 \text{ nm}$ ($=\lambda_{\text{ref}}$). Although the AE30 has a 571 nm channel that is even closer to 532 nm than the 590 nm

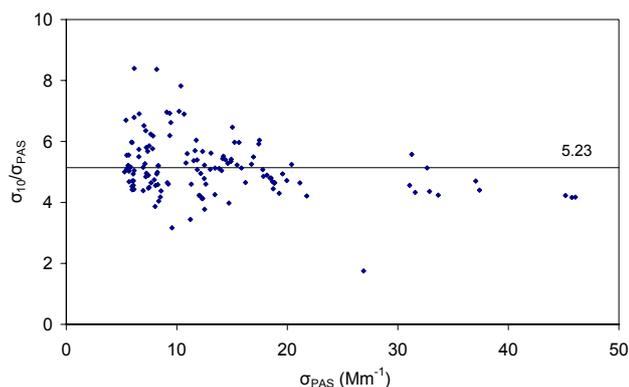


Fig. 6. Experimental determination of the multiple scattering factor C_{532} for the Aethalometer according to Eq. (10). The solid line represents the arithmetic mean ($=5.23$) of the ratio of σ_{10} and σ_{PAS} at 532 nm.

channel, it had to be discarded, since for unknown reasons it was consistently too low by about 20%.

3.2.1 Multiple scattering and loading correction

Following Eq. (10) the multiple scattering correction factor $C_{532}=5.23\pm 0.17$ was determined from the arithmetic mean (95% confidence level of the mean) of the ratios of σ_{10} and σ_{PAS} (see Fig. 6), where again we limited σ_{PAS} to values larger than 4 Mm^{-1} . The multiple scattering correction is by far the most important effect when inferring σ_{aeth} from σ_{ATN} .

The effect of filter loading on Aethalometer sensitivity is depicted in Fig. 7. Each measurement cycle of the Aethalometer begins with an acclimatization phase during which a pristine spot of the filter tape is put into place and the measured ATN (triangles) is defined as 0, i.e., the light intensity (I) measured through the sample spot is set equal to the intensity transmitted through a clean reference spot (I_0 , see Eq. 5). With continuing exposure to the sample flow increasing amounts of (absorbing) aerosol deposit onto the filter spot and the resulting “darkening” of the filter progressively increases the light attenuation ATN (open triangles). At a predefined ATN value of ~ 75 ($\lambda=590\text{ nm}$) the filter tape is automatically forwarded to expose a new pristine filter spot (ATN ~ 0) to the sample flow and the cycle starts again. Figure 7 shows a time series of five consecutive filter changes. For each time layer, we calculated the Aethalometer sensitivity (R_{meas} ; solid diamonds) from Eq. (12) using $C_{532}=5.23$. Fitting the numerical expression of the sensitivity ($R(ATN)$; see Eq. 11) to R_{meas} yields the fit parameter $f=1.20$, where we set $R(ATN)=1$ for $ATN<10$ to be consistent with the assumption adopted for determining C_{532} from Eq. (10). For comparison, we also plotted the fit based on the more rigorously derived form of the loading correction presented by A2005 (dashed line; see their Eq. 27) that shows a very similar result. As seen in Fig. 7, the loading effect accounts for a maximum sensitivity reduction of about 20% at 532 nm. The

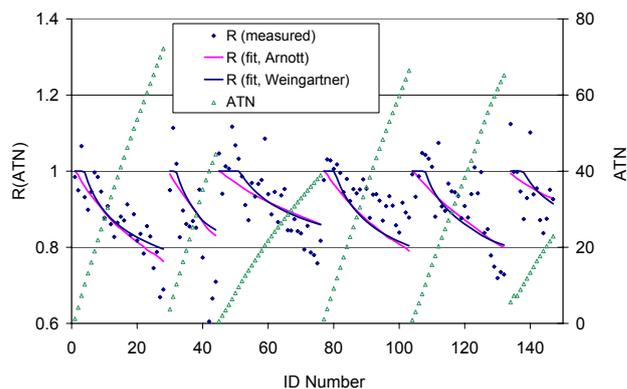


Fig. 7. Illustration of the effect filter loading (ATN, open triangles) on Aethalometer sensitivity (R) for the 590 nm channel converted to 532 nm (data points were numbered consecutively). The measured sensitivity (solid diamonds) was fitted according to the expressions provided by Weingartner et al. (2003) (blue line) and Arnott et al. (2005) (magenta line).

poor correlation coefficient between data and fit ($R^2\sim 0.5$) is a result of the relatively small effect of filter loading ($<20\%$ at 532 nm) compared to the multiple scattering correction factor of 5.23. Hence, small fluctuations in C_{532} may obfuscate the filter loading effect. As discussed below possible culprits for such fluctuations are instrument instabilities and sensitivities to RH , ω_0 , and gaseous adsorption.

With $C_{532}=5.23$ we can now use Eqs. (23) and (24) to find C_λ and hence use Eqs. (9) and (11) to convert σ_{ATN} to σ_{aeth} for each Aethalometer channel (as mentioned above, $f=1.20$ is independent of λ) provided the Ångström exponents for absorption (α_a) is known. However, since α_a is not known, we use the following iterative procedure: 1) use α_{ATN} (from Aethalometer data) as first approximation for α_a , 2) calculate $\sigma_{aeth,\lambda}$ from Eqs. (9) and (11) and (23), 3) derive a refined α_a value based on $\sigma_{aeth,\lambda}$ and 4) repeat steps 2 and 3 until $\sigma_{aeth,\lambda}$ converges. During the SMOCC campaign, α_{ATN} was typically around 1.5. Applying the above procedure we find $\alpha_a=1.82, 1.91, 1.94$ and 1.95 for iterations one through four, i.e., no more than four iterations are required to achieve convergence. It is noteworthy that $\alpha_a=1.95$ is consistent with the values of 1.8 to 1.9 and 2 reported by Schnaiter et al. (2005) and Kirchstetter et al. (2004) for biomass burning particles, respectively. For comparison, using $m_s=0$ as suggested by W2003 ($m_s=0$) yields average α_a values of about 1.5. Hence, this can be interpreted as support for the m_s factors provided by A2005. We note that the values near 1.5 that were also reported by Schnaiter et al. (2005) are not considered relevant here, since they correspond to particle size distributions with unrealistically large count median diameters ($\sim 350\text{ nm}$; here: $CMD<200\text{ nm}$ (Rissler et al., 2006)). Additional laboratory calibrations should be performed to resolve the apparent discrepancies in the m_s calibration factors. In the following we will utilize the calibration factors by A2005 to

estimate the spectral dependence of C (see Table 1).

With $\alpha_a=1.95$ we find from Eq. (24) that $A=-0.1178$ and $B=1.982$ and with Eq. (23) we can write

$$C_\lambda = C_{532} \frac{\lambda^{-0.1178 \ln(\lambda/\text{nm})+1.982}}{532 \text{ nm}^{-0.1178 \ln(532)+1.982}}, \quad (28)$$

i.e., $C_{450}=4.79$, $C_{590}=5.50$, $C_{615}=5.61$, $C_{660}=5.80$, $C_{880}=6.54$ and $C_{950}=6.73$. Hence, for the five Aethalometer channels between 450 to 660 nm, C_λ increases by 10.9, 15.9 and 21.6% for $\alpha_a=1.0$, 1.5 and 2.0, respectively. Consequently, for the SMOCC campaign with $\lambda_{\text{ref}}=532$ nm, assuming a λ -independent C ($=C_{532}$) value for the Aethalometer channels between 450 and 660 nm introduces an error in σ_{aeth} of less than ± 5.5 , ± 7.9 and $\pm 11.2\%$ for $\alpha_a=1.0$, 1.5 and 2.0, respectively. For the worst case (950 nm), C is expected to be 9.0, 19.0 and 31.5% larger than C_{532} for $\alpha_a=1$, 1.5 and 2, respectively, i.e., except for possibly $\alpha_a=1.0$, the wavelength dependence of C can not be neglected. This seems to be in contradiction to W2003 who concluded from the small (up to 10%) increase in C_λ from 450 to 660 nm for soot particles (with $\alpha_a \sim 1.0$), that the spectral dependence of C is negligible. As seen from Table 1 their measurements are consistent with A2005 (10.9% increase in C for this case), but their conclusion is limited to the spectral range between 450 and 660 nm as discussed above.

In summary, we argue that for the SMOCC data f is equal to 1.2 independent of wavelength, while C depends on wavelength and, consequently, the best Aethalometer accuracy is obtained, if C_λ is calculated from Eq. (28). On the other hand, assuming C to be independent of wavelength ($C_\lambda \cong C_{532}$) for the five AE30 channels between 450 and 660 nm does not introduce systematic errors larger than $\pm 11\%$, a bias that is much smaller than the calibration uncertainty ($\sim 20\%$; see Sect. 3.2.5). However, for the 850 nm and 950 nm channels Eq. (28) should be used. This restriction can be relaxed for sample aerosol with $\alpha_a \cong 1.0$, where $C_\lambda \cong C_{532}$ does not introduce biases larger than 9%.

3.2.2 Dependence on sampling period

Since the Aethalometer response is known to depend on aerosol properties and hence on sampling location (Petzold et al., 1997; Arnott et al., 2005), it is conceivable that the correction factors C and f varied with pollution level and sampling period. Using $C_{532}=5.23$ and $f=1.20$, Figs. 8a and b show the ratio of σ_{aeth} (at 532 nm) and σ_{PAS} as a function of the pollution level (indicated by σ_{PAS}) for both the dry and transition period of the SMOCC campaign, respectively, where the seasonal mean values of 0.943 and 1.034, respectively, are indicated by horizontal lines. While there is no systematic dependence of $\sigma_{\text{aeth}}/\sigma_{\text{PAS}}$ on σ_{PAS} for the transition period, there is a small negative trend for the dry period, which will result in a 13% difference in slopes derived from the linear regression of σ_{aeth} and σ_{PAS} (data not shown) given by $\sigma_{\text{aeth}}=0.87 \sigma_{\text{PAS}} (\text{Mm}^{-1})+0.98 \text{ Mm}^{-1}$ ($R^2=0.91$)

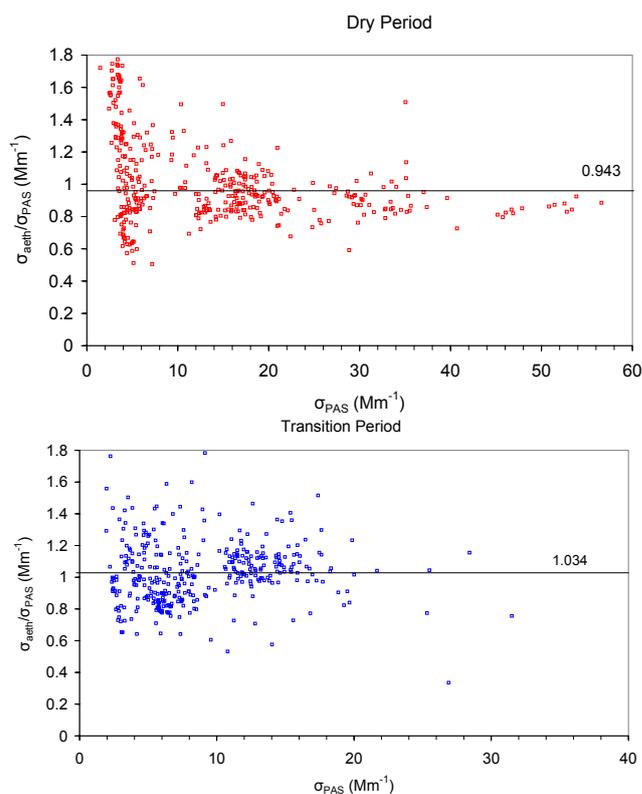


Fig. 8. Dependence of the normalized corrected Aethalometer data on pollution level (represented by σ_{PAS}) and sampling period, namely the dry (a) and transition period (b).

and $\sigma_{\text{aeth}}=1.00 \sigma_{\text{PAS}} (\text{Mm}^{-1})+0.49 \text{ Mm}^{-1}$ ($R^2=0.73$) for the dry and transition period, respectively. The larger slope (and arithmetic mean of $\sigma_{\text{aeth}}/\sigma_{\text{PAS}}$) for the transition period is consistent with the previously discussed enhanced aerosol mass bias (8% and 15% for the dry and transition period, respectively) induced by the difference in inlet cut-off diameters (10 μm versus 1.5 μm). However, despite these small differences we conclude that there is no systematically significant dependence of the Aethalometer correction factors on sampling period. Thus, unless stated otherwise, we will henceforth not distinguish between dry and transition period.

For the wet season, it was impossible to calibrate the Aethalometer and PSAP, mainly due to the poor signal-to-noise ratio and the unavoidable small drifts in zero offset of the PAS. Hence, for lack of a better alternative, we recommend to apply the correction factors derived for the dry and transition period also to the Aethalometer and PSAP data of the wet period.

3.2.3 Dependence on relative humidity

As mentioned above, while the PAS was operated under dry conditions ($RH < 45\%$), the sample air supplied to the Aethalometer was not actively dried, i.e., it closely

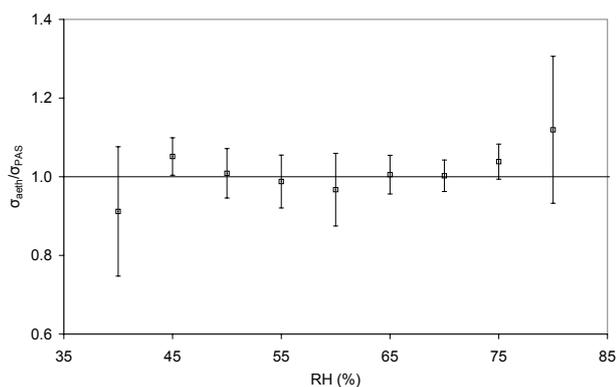


Fig. 9. The effect of relative humidity (RH) on Aethalometer performance. Depicted are the mean and 95% confidence level of the ratios of ambient and dry absorption coefficients as measured by the Aethalometer (σ_{aeth}) and the photoacoustic spectrometer (σ_{PAS}), respectively.

approximated ambient conditions. Figure 9 depicts the Aethalometer-based ambient absorption coefficient (σ_{aeth}) normalized to dry absorption (σ_{PAS}) as a function of RH . For each RH segment, the mean and 95% confidence level of the mean was calculated. It is evident that there is no statistically significant dependence of σ_{aeth} on RH at least for RH between 40 and 80%.

3.2.4 Dependence on single scattering albedo ω_0

The effect of ω_0 on the Aethalometer signal can be assessed based on Eq. (17). Laboratory studies by W2003 and A2005 reported m_s values of ~ 0.008 and 0.055 (at 550 nm), respectively, for purely scattering aerosol (i.e., up to 5.5% of aerosol light scattering is erroneously interpreted as absorption). During the dry and transition period of the SMOCC campaign, ω_0 was approximately constant at 0.92 ± 0.02 (Chand et al., 2006). Using Eq. (17) with $m_s = 0.055$ as an estimated upper limit of the aerosol scattering effect, we can attribute up to $13.0 \pm 3.5\%$ of the observed multiple scattering correction $C_{532} (=5.23)$ to aerosol light scattering, i.e., the multiple scattering factor corrected for aerosol scattering is given by $C_{532}^* = 4.55$. Hence, compared to multiple scattering within the filter matrix ($C_{532}^* = 4.55$) the effect due to the scattering properties of the aerosol is relatively small. While for urban pollution, this effect can be expected to be even smaller (smaller ω_0), the scattering contribution for desert dust and (maritime) background is likely to be larger. Using $\omega_{0,550} = 0.965$ as a typical average value for background air and desert dust (Dubovik et al., 2002). C_{532}^* would have to be increased by about 40% to account for aerosol scattering effects, i.e. $C_{532} \sim 6.4$.

Finally, we note that, since the small variability in ω_0 (± 0.02) during SMOCC translates into a relatively small effect on C_{532} (3.5%), it would have been impossible to distinguish the aerosol scattering (m_s) from the multiple scattering

effect of the filter substrate, i.e., it is impossible to derive a value for m_s from the SMOCC data.

3.2.5 Gaseous adsorption onto the filter

Gaseous adsorption onto quartz filters is a well-known phenomenon that potentially enhances the multiple scattering effect of the filter, and hence introduces a positive bias in C (Kirchstetter et al., 2001). To our knowledge, there has been no previous study of this phenomenon for either the PSAP or the Aethalometer.

The intuitive approach for an investigation of this effect is to look for systematically enhanced C values for large pollution levels. However, since each Aethalometer measurement cycle begins with an acclimatization phase, which exposes the (initially) clean filter spot to ambient air without taking data, and references the measured attenuation to the zero value obtained during this acclimatization phase, a potential dependence of C on pollution level is eliminated. This explains why we found no statistically significant dependence of C_{532} on sampling period despite the substantially higher pollution levels during the dry period with an average σ_a (550 nm) of 22.9 and 7.5 Mm^{-1} for the dry and transition period, respectively (see Fig. 8).

However, if gaseous adsorption introduces a bias into C it can be detected according to the following rationale. Let us assume that at time t_0 the filter is in equilibrium with the gas phase, i.e., there is no net transport of gas molecules to or from the filter. If the pollution level changes at time t_1 , there will be a net transport of gas molecules to or from the filter depending on whether the pollution level increases or decreases, respectively. Furthermore, if gaseous adsorption is present and has an effect on C and if the relaxation time for adsorption/desorption is smaller than the averaging time of the Aethalometer (here 15 min) one would expect a systematic dependence of the relative gradient of C on the gradient in pollution level. Expressing the gradient of C (here C_{532}) at time layer i as

$$\frac{\Delta C}{C} = \frac{C^{i+1} - C^i}{(C^{i+1} + C^i)/2} \quad (29)$$

and using the gradient in CO as proxy for changes in pollution level

$$\frac{\Delta \text{CO}}{\text{CO}} = \frac{\text{CO}^{i+1} - \text{CO}^i}{(\text{CO}^{i+1} + \text{CO}^i)/2} \quad (30)$$

we found no correlation ($R^2 < 0.1$) between the relative gradients in C and CO. The same result was found, if NO_2 instead of CO was used as proxy for the pollution level. Performing this analysis also for the PSAP yielded the same result ($R^2 < 0.04$). It is important to note that we do not suggest that NO_2 or CO actually adsorb to the filter; these components only serve as a proxy for pollution events driven by photochemistry and/or biomass burning. Although we can

not rule out the possibility that there are adsorbing gaseous components that do not correlate well with CO or NO₂, this analysis suggests that neither the Aethalometer nor the PSAP suffer from significant measurement artifacts due to adsorption of gaseous components.

3.2.6 Accuracy

The accuracy of σ_{aeth} is determined by the accuracy of C . As mentioned above, for wavelengths between 450 and 660 nm, C_{λ} can be approximated by C_{532} with an estimate accuracy and precision (95% confidence level) of about 20% and 30% (15-min averages), respectively, except for the 571 nm channel, which was systematically too low. For wavelengths larger than 660 nm (here 880 and 950 nm), this level of accuracy can only be maintained, if C_{532} is converted into C_{λ} using Eq. (28). The Aethalometer accuracy is lower than the PSAP accuracy to account for additional uncertainties due to wavelength conversion (uncertainties in α_a) and the use of different inlets.

4 Discussion

4.1 PSAP

The PSAP was used in numerous field studies (Reid et al., 1998; Anderson et al., 1999; Mader et al., 2002; Wex et al., 2002; Arnott et al., 2003; Guyon et al., 2003a, b). Most of these studies accounted for instrument artifacts due to flow rate, spot size and aerosol scattering using the calibration factors by B1999. For Amazonian aerosol, we found that the Bond correction did not account for a bias that scaled well with RH and T (see Figs. 5a and b). As seen in Fig. 5c this bias can be accounted for by using a RH dependent correction factor (see Eq. 27). In this section we compare these findings to other PSAP calibration studies.

V2005a has essentially confirmed the Bond correction for external mixtures of kerosene soot and ammonium sulfate particles. However, the observed deviations for pure soot (from a kerosene lamp) and for purely scattering particles resulted in the derivation of a new (ω_0 dependent) loading correction. Since the Bond calibration was performed with spherical particles (internal mixtures of nigrosin and ammonium sulfate), inconsistencies for fractal-like soot agglomerates as described by V2005a are conceivable. Unfortunately, V2005a was unable to derive a “unified” correction scheme that would be applicable to all types of aerosols used (ammonium sulfate, PSL, soot, and external mixtures thereof). During the ~ 1.5 day period of their outdoor experiment in Reno, NV, V2005a reported PSAP absorption coefficients that were by 16% and 22% higher than the reference absorption at 530 nm where using their own correction scheme (1- λ PSAP correction parameters) and the Bond correction, respectively. This compares well with the +18% bias of $\sigma_{\text{PSAP,Bond}}$ reported in this study for $RH > 30\%$ (Fig. 5a). To our know-

ledge, there are only two more PSAP field calibrations with a true in-situ reference method such as the difference method or the PAS. The study by Reid et al. (1998) did not account for PSAP artifacts, since it was performed prior to B1999 and Arnott et al. (2003) found that $\sigma_{\text{PSAP,Bond}}$ was by a factor of 1.61 larger than σ_{PAS} (532 nm) for rural aerosols from the North Central Oklahoma.

None of the previous PSAP calibrations reports a systematic dependence of the PSAP performance on RH and/or T . On the other hand, there is evidence for a RH sensitivity of the PSAP. For instance, Arnott et al. (2003) reported an erratic response of the PSAP for rapidly changing RH (also seen by us) and Guyon et al. (2004) had to discard PSAP data, if RH exceeded 92%. Changes in RH may affect the amount of water that condenses into the cavities of the filter matrix. As a consequence the optical properties of the filter may change and hence filter-based absorption techniques may depend on RH . In addition, hygroscopic aerosol growth may “truly” enhance aerosol light absorption due to the optical interaction between aerosol core and coating (Fuller et al., 1999), but for ambient aerosol absorption enhancement factors larger than about 1.1 are difficult to justify at least for $RH < 80\%$ (Redemann et al., 2001). As seen from Figs. 5a and b the normalized PSAP response ($\sigma_{\text{PSAP,Bond}}/\sigma_{\text{PAS}}$) increased by a factor of 2 for an increase in RH from 20 to 30% and/or an increase in T from 24 to 26°C, respectively. Hence, we conclude that the observed bias is an instrument artifact of either the PSAP or the PAS. As mentioned above, there is both theoretical and experimental evidence for the absence of an RH sensitivity of the PAS for $RH < 80\%$ (Raspert et al., 2001; Arnott et al., 2003) and the moderate dependence of the PAS signal on T is accounted for by repeated acoustic calibrations (every 10 min) as described above. In addition, comparison of the Aethalometer and the PAS has not shown any RH or T dependent bias. Hence, we attribute the observed bias of $\sigma_{\text{PSAP,Bond}}/\sigma_{\text{PAS}}$ to a systematic error of the PSAP.

As mentioned above both RH and T displayed a pronounced diel variation such that most of the low RH and T values were encountered during nighttime sampling. The resulting correlation of RH and T makes it impossible to conclusively decide whether the observed PSAP bias is due to RH or T or both. The significant scatter in both Figs. 5a and b suggests that there is possibly a sensitivity of the PSAP to both RH and T or may be even to another – as yet unidentified – parameter. As mentioned above $\sigma_{\text{PSAP,Bond}}/\sigma_{\text{PAS}}$ did not show any correlation with filter loading (for transmittance > 0.5 , i.e., $\text{ATN} < 70$) or single scattering albedo (for $0.85 < \omega_{0,550} < 0.95$). Furthermore, laboratory measurements indicate that $\sigma_{\text{PSAP,Bond}}$ does not depend on σ_{PAS} or σ_s (V2005a). To our knowledge a potential PSAP sensitivity to RH and T has not been systematically investigated yet. If RH is the culprit, the drastic change near $RH = 20\%$ may be a result of a wettability threshold of the PSAP filter near $RH = 20\%$. On the other hand, it is also conceivable that there

is a temperature sensitivity of the PSAP electronics and/or optics that results in the observed drift of the PSAP. This issue should be explored further in a laboratory study under controlled RH and T conditions. For the purposes of this study, it is sufficient to adjust the Bond-corrected PSAP data with the factor K_{RH} (Eq. 27). The resulting PSAP-derived absorption coefficient (σ_{PSAP}) showed excellent agreement ($<6\%$) and correlation ($R^2=0.954$) with the reference absorption coefficient (σ_{PAS}) (see Fig. 5c).

In summary, we conclude that the Bond correction is generally quite adequate for most ambient aerosols ($\sim 25\%$ accuracy) with three possible exceptions: (1) purely fractal-like agglomerates, (2) purely scattering particles (i.e., ω_0 close to unity) and (3) varying RH and/or T conditions (for $RH < 30\%$ and $T < 25^\circ\text{C}$; see Figs. 5a and b). In this study nighttime PSAP data had to be corrected by a RH -dependent correction factor.

4.2 Aethalometer

For Amazonian aerosol, the Aethalometer calibration revealed a multiple scattering correction factor of $C_{532}=5.23\pm 1.04$ (or $C_{532}^*=4.55\pm 0.91$, if corrected for aerosol scattering effects according to A2005, i.e., 13% of C_{532} can be attributed to aerosol scattering effects) and a much less significant bias due to filter loading ($0.76 < R \leq 1$ at 532 nm). Previous calibration studies reported C values clustering around 2 or 4 depending on the type of challenge aerosol (W2003, A2005). For pure candle light soot particles and external mixtures of soot and ammonium sulfate, A2005 found C^* values between 1.8 and 2.2 (depending on wavelength). On the other hand, they reported $C_{521}^*=3.7$ for ambient (urban) aerosols. Similarly, the laboratory study by W2003 found $C_{532}^*=2.14$ (here $C=C^*$, since, in contrast to A2005, W2003 considered the dependence of the Aethalometer performance on aerosol light scattering as negligible) for both pure soot (Diesel and PALAS) and external soot mixtures with ammonium sulfate (independent of wavelength). On the other hand, for soot (Diesel and PALAS) particles coated with organic carbon (internally mixed aerosol) their C^* value increased to 3.6 which is very similar to 3.7 and 4.55 ± 0.91 as reported for ambient aerosol by A2005 and by the present study, respectively. We remind the reader that the value of 4.55 can be considered an upper limit due to the (unaccounted) positive bias (estimated as <8 and 15% for the dry and transition period, respectively) resulting from the larger cut-off diameter of the Aethalometer inlet.

Both A2005 and W2003 offered possible explanations for the factor of ~ 2 difference in C^* . A2005 hypothesized that variable particle preloading of the filter during the automatically performed filter acclimatization phase prior to any measurement might be responsible for the enhanced C^* value under ambient conditions. However, in light of a maximum loading correction of no more than 30% (~ 550 nm), as

was consistently reported by W2003, A2005 and the present study, a factor of 2 difference in C^* seems hard to justify. On the other hand, W2003 speculated that adsorption of semi-volatile organic gaseous components onto the filter matrix might have artificially enhanced the multiple scattering within the filter matrix. However, as shown in Sect. 3.2.4, we found no indication for gaseous adsorption effects during the SMOCC campaign.

Although we are unable to resolve this issue conclusively, we offer a different explanation for the observed difference in C^* . The significance of the aerosol mixing state for light absorption is well known from Mie theory for coated particles (Bohren and Huffman, 1983). Petzold et al. (1997) argued that, for internal mixtures of black carbon (BC) and some mainly scattering material, the Aethalometer response may be enhanced by up to about a factor of 2 (for BC contents of about 3%) compared to external mixtures (as was the case for the laboratory calibrations by W2003 and A2005). Hence, we suggest that the observed difference in C^* may possibly be a result of attenuation enhancement due to internal mixing. This notion is corroborated by the fact that the Bond correction of the PSAP, that demonstrates good applicability to ambient aerosol, was performed with internal mixtures of nigrosin and ammonium sulfate (V2005a).

In summary, we suggest that in absence of an on-site calibration standard, C_{532}^* values of 2.1 and 4.0 (average of 3.7, 3.6 and 4.55) should be used for pure or external mixtures of soot and internal mixtures of soot, respectively. The additional bias due to aerosol scattering can be taken into account by using Eq. (20) with the m_s factors provided by A2005 (see Table 1). For particles with small absorption Ångström exponents ($\alpha_a \cong 1$), C_λ can be approximated by C_{532} ($<10\%$). For α_a values up to 2 (as found during SMOCC), we found that C_λ can be approximated by C_{532} ($<11\%$) for the five AE30 channels between 450 and 660 nm. However, for the 880 and 950 nm channel, the spectral dependence of C should be accounted for by using Eq. (23). Since the fit coefficients of Eq. (23) depend on ω_0 and α_s , these coefficients are aerosol-specific and therefore they should be calculated for the aerosol under consideration as described here for the SMOCC data ($\omega_{0,550}=0.92$, $\alpha_s=2$).

The filter loading correction factor R depends on attenuation and hence on λ . At the highest loading prior to the automatic filter change (ATN=100, 75 and 40 for 450, 532 and 950 nm) R is 0.76, 0.8 and 0.85 for $\lambda=450$, 532, and 950 nm, respectively, i.e., the measured attenuation coefficient on a pristine filter is 32, 25, and 18% larger than on a fully loaded filter, respectively. This is consistent with the values reported by W2003, i.e., in contrast to C , R (and hence f) does not seem to depend on the mixing state of the sample aerosol. Furthermore, W2003 showed that the shadowing factor f (see Eq. 11) is related to ω_0 by $f=A(1-\omega_0)+1$, where $A=0.86\pm 0.1$. Using $\omega_0=0.92$ (observed during SMOCC) yields $f=1.07\pm 0.01$. Although this value is somewhat lower than experimentally determined value of 1.2, it

results in a less than 10% bias in σ_{aeth} , if we average over an entire filter cycle. W2003 acknowledged that the reliability of A is limited due to significant scatter in the data and the validity of A is possibly limited to “dark” aerosol, since most of their measurements were performed for $\omega_0 < 0.6$ and the few data points with $\omega_0 > 0.6$ have large error margins. Hence the apparent inconsistency between f and ω_0 is not surprising especially, if we also consider that f not only depends on ω_0 but on how deep the aerosol is embedded into the filter matrix (A2005), which may depend on particle size and morphology as well as on sampling flow rate.

In contrast to the PSAP, the Aethalometer shows no dependence on RH (or T) within the experimental uncertainty at least for RH between 40 and 80% (Fig. 9). Similar to artifacts due to gaseous adsorption, RH related Aethalometer artifacts should be at least partially accounted for by the filter acclimatization phase. Thus, one might conclude from Fig. 9 that there is no hygroscopic absorption enhancement. This is consistent with the previously mentioned theoretical predictions of Redemann et al. (2001), who estimated the hygroscopic absorption enhancement factor at $\lambda=550$ nm as ~ 1.1 at $RH=80\%$ for sulphuric acid coated soot particles with a realistic lognormal size distribution (geometric mean diameter and standard deviation of $0.12 \mu\text{m}$ and 1.5 , respectively). In light of the small hygroscopic diameter growth factor of less than 1.08 ($RH=80\%$) for Amazonian aerosols (Rissler et al., 2006), we consider this absorption enhancement factor (1.1) to be an upper limit for our study. However, we add as an important caveat that the Aethalometer may not be capable at all of accurately measuring the electromagnetic focusing effect of absorbing particles enclosed by a liquid coating, since the shape (and hence the optical properties) of the partially liquid particle is expected to change upon deposition onto a filter substrate.

Finally, in Figs. 10a and b we compare 1 h averages of PSAP and Aethalometer (adjusted to 532 nm) data for the dry and transition period, respectively. It is evident that the instruments are well correlated for both periods (dry: $R^2=0.88$; transition: $R^2=0.90$) and, forcing the linear regression line through the origin, σ_{PSAP} is by about 9.8% larger (slope = 1.098 ± 0.047) and 2.5% smaller (slope = 0.975 ± 0.030) than σ_{aeth} for the dry and transition period, respectively. Considering that Fig. 10 represents more than 2 months of data, while the calibration of the PAS and PSAP was based on only 200 h, the agreement of the instruments is quite satisfactory and the slopes agree within the estimated instrument accuracies. On the other hand, the correlation is weaker for $\sigma_a > 40 \text{ Mm}^{-1}$. We attribute this to the scarcity of PAS data for this absorption range as seen from Figs. 5c and 8. Hence, we add as a caveat that the reliability of the PSAP and Aethalometer calibration is somewhat weaker for $\sigma_a > 40 \text{ Mm}^{-1}$, although even in this range the agreement between PSAP and Aethalometer is better than 25%, the estimated 2σ level based on the instrument accuracies.

5 Conclusions

A 1- λ PSAP (Particle Soot Absorption Photometer, 565 nm) and a 7- λ AE30 Aethalometer (450–950 nm) were compared to a PAS (photoacoustic spectrometer, 532 nm) based on 200 h of collocated ambient sampling at a rural site in the Amazon Basin during the dry and wet-to-dry transition period of the LBA-SMOCC campaign in 2002. The data are heavily influenced by biomass burning events. To ensure data quality we verified the PAS accuracy of 10% in the field following the calibration procedure described by Arnott et al. (2000) using NO_2 as calibration gas.

The calibration of the PSAP with the PAS essentially confirmed the Bond correction (B1999), except for a previously not reported bias that correlated well with RH and/or T for low RH (20 to 30%) and T values (24 to 26°C). Although the data presented here does not provide conclusive evidence that the PSAP is sensitive to RH and/or T , we were able to account for an observed systematic bias in the PSAP data using a RH dependent correction factor. The manufacturer-provided loading correction was found adequate for transmissions down to 0.5 ($\text{ATN} < 70$). For the limited range of single scattering albedos encountered here ($0.85 < \omega_{0,550} < 0.95$), no ω_0 sensitivity was observed. Based on field calibrations we estimated the accuracy (95% confidence level) of the Bond-corrected PSAP data as about 25%. With the additional on-site PAS calibration, the value improved to about 15%.

For Amazonian haze particles, the multiple scattering correction factor of the Aethalometer at the reference wavelength of 532 nm was $C_{532} = 5.23 \pm 1.05$ or, if aerosol scattering effects are subtracted, $C_{532}^* = 4.55 \pm 0.91$. The loading correction ($0.7 < R < 1$) was adequately described by Eq. (11) using a shadowing factor of $f=1.2$. Based on the limited available information in the literature, we argued that the shadowing factor $f=1.20$ is independent of wavelength (bias $< 10\%$) and the wavelength dependence of C can be parameterized using the data by A2005, if the single scattering albedo ω_0 (at a reference wavelength) and the scattering Ångström exponent α_s are known (Eqs. 23 and 24). This parameterization showed that C_λ can be approximated by C_{532} to better than 11%, if either the absorption Ångström exponent α_a is close to unity (i.e. soot-dominated aerosol) or the spectral range is limited to between 450 to 660 nm. For the Amazonian haze aerosol with α_a near 1.95, C_λ was calculated for each of the seven Aethalometer channels based on Eqs. (23) and (24). Not accounting for the spectral dependence of C would lead to a positive bias of about 30% in the 950 nm channel and a substantially reduced α_a value of about 1.5. We also note that, for unknown reasons, the 571 nm channel of the AE30 Aethalometer was consistently by about 20% too low. For all other channels, the (2σ) accuracy was estimated as 20%.

We found no sensitivity of the PSAP and Aethalometer to gaseous adsorption onto the filter matrix. In addition, except for the aforementioned instrument artifact of the PSAP that correlated well with RH (and T), we found

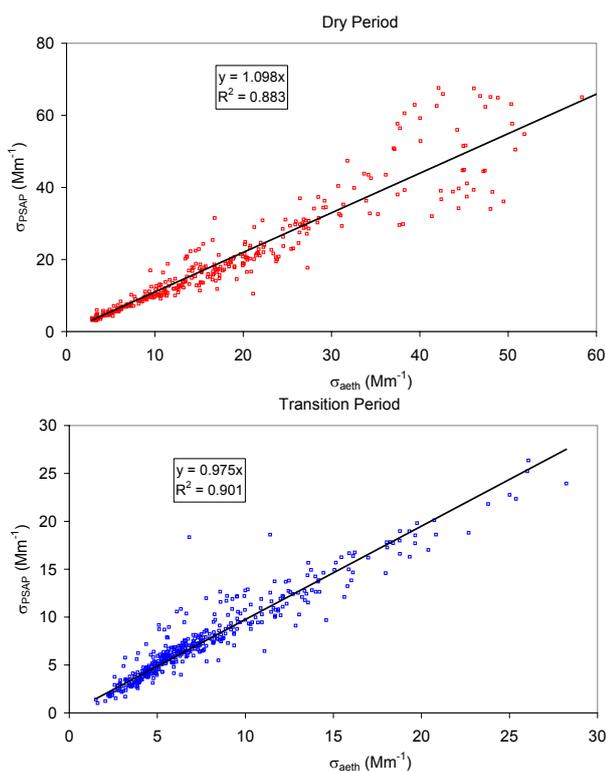


Fig. 10. Comparison of the absorption coefficients (at 532) measured by PSAP and Aethalometer for both the dry (a) and transition period (b).

also no dependence of aerosol absorption (σ_a) on RH . While the absence of an hygroscopic absorption enhancement for $RH \leq 80\%$ is consistent with theoretical predictions (Redemann et al., 2001), it is questionable whether humidified particles deposited onto a filter substrate display the same optical properties as in the suspended state.

This study shows that, while laboratory calibration experiments are useful, on-site calibrations of the PSAP and Aethalometer are required for ambient measurements to ensure data quality. Although both PSAP and Aethalometer are based on the integrating-plate method, the conversion of the measured attenuation (σ_{ATN}) into absorption (σ_a) requires different correction parameters due to the different filter types used. If an on-site calibration cannot be provided, we offer the following suggestions for retrieving absorption coefficients from PSAP or Aethalometer data:

- 1) For the 1- λ PSAP, the Bond correction (Eq. 25) can be applied with an expected uncertainty of about 25% (2σ) and RH and T should be kept constant to avoid potential RH and T induced biases.
- 2) For the 7- λ Aethalometer, the situation is more complicated. We suggest a multiple scattering correction factor C_{532}^* of either 2.1 for pure and externally mixed soot

(e.g., near a combustion source) or 4.0 for internally mixed aerosol (e.g., for aged ambient aerosol). If concomitant measurements of spectrally resolved aerosol scattering are available, Eq. (20) can be used to determine C_{532} . Otherwise, the additional bias due to aerosol scattering can be approximately accounted for by increasing the C_{532}^* values by 10% for non-background air (10% corresponds to $\omega_{0,550} \sim 0.90$) to yield C_{532} values of 2.3 and 4.4 for externally and internally mixed aerosol, respectively. For background and desert dust regions an increase of about 40% (corresponds to $\omega_{0,550} \sim 0.965$), i.e., C_{532} values of 2.9 or 5.6, may be more realistic. If the absorbing component of the sample aerosol is mainly due to soot from internal combustion engines, α_a is close to 1 (Kirchstetter et al., 2004), i.e., $C_\lambda \cong C_{532}$ ($< 11\%$ bias) for $450 \text{ nm} < \lambda < 950 \text{ nm}$. On the other hand, if the absorbing aerosol component mainly results from biomass combustion, α_a is closer to 2 (Kirchstetter et al., 2004) and at least C_{880} and C_{950} should be calculated from Eq. (28). The loading correction R is given by Eq. (11) with $f=1.2$ for $\omega_{0,532} \sim 0.9$ and for $\omega_{0,532}$ smaller than about 0.85, f can be estimated from $f=0.86(1-\omega_{0,532})+1$ as recommended by W2003. Now Eq. (9) can be used to yield absorption coefficients with an estimated (2σ) accuracy of about 25%.

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