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FRACTIONAL SCALING ANALYSIS APPLIED TO SCALE A PRESSURIZER TEST FACILITY

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ABSTRACT

Local Scaling and Fractional Scaling Analysis (FSA) have been applied at the system level to surge flow transients in a pressurizer with the purpose of sizing a 1/100 volume scale test facility. It was shown that the relation between pressure and volume displacement rates is analogous to that of generalized "*effort*" and "*flow*" in interdisciplinary analysis of complex systems. At the component level, a relation between pressurizer liquid volume and the volume displacement rates is obtained. Properly scaled outsurge transients in a full scale pressurizer have identical pressure and liquid volume histories as compared to a properly designed 1/100 volume experimental facility. FSA allows the rank of the processes *quantitatively* and thereby *objectively* in the order of their importance. For the case in analysis, surge flow dominates all other volume displacement rates (the second in importance was due to the heater). The dominant agent of pressure change in the pressurizer is that due to maximum surge flow. Likewise, the heating power of the experiment, necessary to control the specified out-surge flow transient, is calculated assuming equality in the second largest agent of change, which is that due to heating.

K<u>eywords</u>: Fractional Scaling, Fractional Rate of Change, Fractional Change Metric, Hierarchy of Complex Systems, Scale Distortion, Information Synthesis, Holistic Scaling Approach, Pressurizer, Similarity.

1. INTRODUCTION

Background

This paper presents an application of Fractional Scale Analysis (Zuber et al., 2005), (Wulff et al., 2005) to the pressurizer dome in the integral vessel of the International Reactor Innovative and Secure (IRIS) (Carelli, et al., 2004), (Barroso et al., 2004). The pressurizer dome is treated here as a *system* with two *components*: the vapor and the liquid volumes. The system is modeled from a holistic point of view and scaling is carried out at the *system level*. The system pressure is the generalized "effort" affecting simultaneously all components and it accounts not only for the fluid inventory but, more importantly, for the energy stored in the compressed fluids (Wulff et al., 2005). For a two-region pressurizer, the system pressure variation is a consequence of "volume" displacements, or generalized "flows", caused by twelve agents of change. Five of them act inside the vapor volume component and the other seven in the liquid volume. At the component level, the variation of the liquid volume is a consequence of "volume" displacements caused by nineteen types of agents of change. Twelve are derived from pressure variation and the other seven

originated from liquid specific volume. There are <u>two</u> distinct and important system characteristics: the *Fractional Rates of Change* ω (FRC) that *quantifies* the intensity of each transfer process (agent of change) affecting the state variables in terms of what fraction of the variable total change the agent was responsible for. This serves to rank the agent impact on the system (pressurizer). The other is the *Fractional Change Metric* Ω for scaling the fractional change of the system state variables.

Purpose of the work

FSA is applied at system and component levels to find what conditions the (1/100 volume) IRIS Pressurizer Test Facility (IRIS-PTF) should obey to produce the same properly scaled depressurization history *and* liquid volume history as the full size IRIS. For this comparison we have used a computed outsurge transient results from a two region pressurizer model (Botelho et al., 2005a).

2. PRESSURE TRANSIENT

Pressure variation is considered uniform throughout the pressurizer, because the flow between the modules (regions in the pressurizer) is not choked. There are heat losses through the metallic walls of the pressurizer in contact with external ambient air, and some heat losses through the thermal insulation of the pressurizer liquid from remaining primary fluid in the reactor vessel (Barroso et al., 2003). To compensate for heat losses and to control the pressure drop in an outsurge, a heater is provided in the liquid region, which is actuated accordingly from signals of pressure probes (Barroso and Batista, 2004) and (Botelho et al., 2005a). The IRIS pressurizer layout is shown in Figure 1.



Figure 1. The IRIS pressurizer layout

A summary of the governing equations in the liquid and vapor volumes of a pressurizer can be seen in Botelho et al., 2007.

An equation for the pressure can be readily obtained from the total volume constraint combined with mass and energy conservation equations in both volumes as:

$$\frac{d p}{dt} = \frac{\sum_{i=l,v} \left(\mathbf{v}_i \, W_i + \left(\frac{\partial v_i}{\partial h_i}\right)_p \left(\dot{Q}_i - W_i \, h_i\right) \right)}{-\sum_{i=l,v} \left(m_i \left(\frac{\partial v_i}{\partial p}\right)_{h_i} + m_i \mathbf{v}_i \left(\frac{\partial \mathbf{v}_i}{\partial h_i}\right)_p \right)}$$
(1)

2.1. Closing Equations and Initial Conditions

Prior to a transient we assume steady-state conditions and $\dot{m}_{surge} = 0$. On the interface between the vapor and liquid volumes, the mass flow rate is calculated using a formula derived from the kinetic theory of the gases, and the heat transfer coefficient from a Nusselt number correlated for similar pressurizer experiments (Kang and Griffith, 1984).

The heat losses are calculated solving the heat conduction equation on the pressurizer walls. The wall condensation heat transfer is calculated from a Nusselt model (*in* Holman, 1989).

Bubble *terminal velocity*, as calculated by an experimental correlation (Wilson et al., 1961) or Zuber simpler formula (*in* Whalley, 1996), is assumed for all bubbles reaching the interface and used to calculate the flashing mass flow rate.

For the *rainout*, the velocity reached at the interface by an *average* falling drop immersed in the vapor fluid is used to calculate its mass flow rate.

Isentropic Compressibility

Eq. (1) can be rewritten in the following form

$$\frac{d p}{dt} = \frac{1}{VK_{s, sys}} \sum_{i=l,v} \left(\mathbf{v}_i \ W_i + \left(\frac{\partial v_i}{\partial h_i}\right)_p \left(\dot{Q}_i - W_i \ h_i\right) \right) \qquad = \frac{\sum_{i=l,v} V_j}{VK_{s, sys}} = \sum_{all agents of change j} \Phi_j, \tag{2}$$

- .

to define the individual generalized rates of volume change, \dot{V}_j , and the individual rates of pressure change, Φ_j for all agents of change. The denominator $K_{s,sys}$ is the isentropic compressibility of the global system, i.e. the sum of the volume weighted isentropic compressibility of the *aggregate sub-volumes*

$$K_{s,sys} = \sum_{i=l,v} \frac{V_i}{V} (\kappa_s)_i$$
(3)

where:

$$\left(\kappa_{s}\right)_{i} = \frac{-m_{i}\left(\frac{\partial \mathbf{v}_{i}}{\partial p}\right)_{h_{i}} - m_{i}\mathbf{v}_{i}\left(\frac{\partial \mathbf{v}_{i}}{\partial h_{i}}\right)_{p}}{V_{i}}$$
(4)

The terms in the summation bracket in Eq. (2) represent all the possible agents of change affecting the pressurizer pressure during the entire outsurge transient. These terms are functions of the mass and energy transfer terms between the two modules of the pressurizer (vapor and liquid volumes), INAC 2007, Santos, SP, Brazil.

the mass flow rate crossing the surge orifices, the energy provided by the heaters, and losses through the pressurizer walls.

Equation (2) is used to define the individual rates of volume dilatation or contraction, \dot{V}_j in the vapor and liquid volumes, such that

$$\sum_{all j} \dot{\mathbf{V}}_{j} = \sum_{i=l, \sqrt{\mathbf{V}_{i}}} \left(\mathbf{v}_{i} \, W_{i} + \left(\frac{\partial v_{i}}{\partial h_{i}} \right)_{p} \left(\dot{Q}_{i} - W_{i} \, h_{i} \right) \right)$$
(5)

The individual rates of volume dilatation or contraction, \dot{V}_i , for each local phenomenon: surge flow, flashing, wall condensation, rainout, heat balances, and mass transfer terms through the interface between vapor and liquid are:

$$\dot{\mathbf{V}}_{0} = \dot{\mathbf{V}}_{surge} = \left(\mathbf{v}_{l} - \left(\frac{\partial \mathbf{v}_{l}}{\partial h_{l}}\right)_{p} \left(h_{l} - h_{surge}\right)\right) \dot{m}_{surge}$$
(6);
$$\dot{\mathbf{V}}_{1} = \dot{\mathbf{V}}_{l,WC} = \left(\mathbf{v}_{l} + \left(\frac{\partial \mathbf{v}_{l}}{\partial h_{l}}\right)_{p} \left(h_{f} - h_{l}\right)\right) W_{WC}$$
(7)

$$\dot{\mathbf{V}}_{2} = \dot{\mathbf{V}}_{l,RO} = \left(\mathbf{v}_{l} + \left(\frac{\partial \mathbf{v}_{l}}{\partial h_{l}}\right)_{p} \left(h_{f} - h_{l}\right)\right) W_{RO}$$
(8);
$$\dot{\mathbf{V}}_{3} = \dot{\mathbf{V}}_{l,FL} = -\left(\mathbf{v}_{l} + \left(\frac{\partial \mathbf{v}_{l}}{\partial h_{l}}\right)_{p} \left(h_{g} - h_{l}\right)\right) W_{FL}$$
(9)

$$\dot{\mathbf{V}}_{4} = \dot{\mathbf{V}}_{l,\dot{\mathcal{Q}}_{l}} = \left(\frac{\partial \mathbf{v}_{l}}{\partial h_{l}}\right)_{p} \left(\dot{\mathcal{Q}}_{lv} + \dot{\mathcal{Q}}_{heat,0} - \dot{\mathcal{Q}}_{wl}\right) \quad (10); \ \dot{\mathbf{V}}_{5} = \dot{\mathbf{V}}_{l,lv} = \left(\mathbf{v}_{l} + \left(\frac{\partial \mathbf{v}_{l}}{\partial h_{l}}\right)_{p} \left(h_{lv} - h_{l}\right)\right) \dot{m}_{lv} \quad (11)$$

$$\dot{\mathbf{V}}_{6} = \dot{\mathbf{V}}_{\mathbf{v},FL} = \left(\mathbf{v}_{\mathbf{v}} + \left(\frac{\partial \mathbf{v}_{\mathbf{v}}}{\partial h_{\mathbf{v}}}\right)_{p} \left(h_{g} - h_{\mathbf{v}}\right)\right) W_{FL}$$
(12);
$$\dot{\mathbf{V}}_{7} = \dot{\mathbf{V}}_{\mathbf{v},WC} = -\left(\mathbf{v}_{\mathbf{v}} + \left(\frac{\partial \mathbf{v}_{\mathbf{v}}}{\partial h_{\mathbf{v}}}\right)_{p} \left(h_{g} - h_{\mathbf{v}}\right)\right) W_{WC}$$
(13)

$$\dot{\mathbf{V}}_{8} = \dot{\mathbf{V}}_{\mathbf{v},RO} = -\left(\mathbf{v}_{\mathbf{v}} - \left(\frac{\partial \mathbf{v}_{\mathbf{v}}}{\partial h_{\mathbf{v}}}\right)_{p} \left(h_{\mathbf{v}} - h_{f}\right)\right) W_{RO} \qquad (14); \ \dot{\mathbf{V}}_{9} = \dot{\mathbf{V}}_{\mathbf{v},\dot{Q}_{\mathbf{v}}} = -\left(\frac{\partial \mathbf{v}_{\mathbf{v}}}{\partial h_{\mathbf{v}}}\right)_{p} \left(\dot{Q}_{Iv} + \dot{Q}_{wv}\right) \tag{15}$$

$$\dot{\mathbf{V}}_{10} = \dot{\mathbf{V}}_{\mathbf{v}, lv} = -\left(\mathbf{v}_{\mathbf{v}} - \left(\frac{\partial \mathbf{v}_{\mathbf{v}}}{\partial h_{\mathbf{v}}}\right)_{p} (h_{\mathbf{v}} - h_{lv})\right) \dot{m}_{lv} \qquad (16); \ \dot{\mathbf{V}}_{11} = \dot{\mathbf{V}}_{l, \dot{Q}_{heat}} = \left(\frac{\partial \mathbf{v}_{l}}{\partial h_{l}}\right)_{p} (\dot{Q}_{heat}) \tag{17}$$

In an outsurge transient, the strongest agent of change is the surge flow. As will be shown in Tables 4 and 5 below, the maximum surge flow produces 1000 times more volume displacement as flashing, which is the local phenomenon that produces more volume displacement. The maximum heating, the next largest agent, which produces 100 times more volume displacement as flashing (and 10 times less than that of maximum surge flow) is responsible to restore the pressurizer pressure to the initial set point.

3. FRACTIONAL SCALING ON THE SYSTEM LEVEL

The normalized pressure, or *fractional change of pressure*, of order one, is defined by

$$0 \le p^* = \frac{p(t) - p_{\min}}{p_{\max} - p_{\min}} = \frac{p(t) - p_{\min}}{\Delta p} \le 1$$
(18)

An scaled equation can be obtained directly from Eq. (2) by normalizing each dilatation rate with the *effective dilatation rate of the aggregate* as in (Zuber et al., 2005), i.e., the magnitudes of the algebraic sum of all (active) dilatation rates

$$\dot{\mathbf{V}}_{j}^{*} = \frac{\dot{\mathbf{V}}_{j}(t)}{\left|\sum_{j=0}^{11} \dot{\mathbf{V}}_{j}\right|_{t=0}} , \quad \left|\sum_{j=0}^{11} \dot{\mathbf{V}}_{j}\right|_{t=0} \neq 0$$
(19)

Equations (18) and (19) are introduced into Eq. (2) that after division by Δp becomes *directly*

$$\frac{d p^*}{dt} = \left| \sum_{j=0}^{11} \omega_j \right|_{t=0} \sum_{j=0}^{11} \frac{\dot{\mathbf{V}}_j^*}{K_{s,sys}^*},\tag{20}$$

where $K_{s, sys}^* = K_{s, sys} / K_{s, sys, 0}$.

Other terms in Eq. (20) are defined below.

Fractional Rate of Change (FRC)

Every agent of change is represented by a <u>fixed</u> Fractional Rate of Change (FRC), $\mathcal{O}_{i,0}$,

$$\boldsymbol{\omega}_{j,0} = \frac{\dot{\mathbf{V}}_{j,0}}{VK_{s,sys,0} \,\Delta p} \tag{21}$$

The FRCs $\omega_{j,0}$, measure the intensity of their respective agents of change. The facility characteristic $VK_{s,sys,0}$ and the operating parameters Δp and $\dot{V}_{j,0}$ are included in the fixed Fractional Rates of Change, $\omega_{j,0}$. FSA states that *Fractional Change Metrics* $\Omega_j = \omega_{j,0}t_{ref}$ must be common to all scaled facilities of any volume size, system elasticity, surge flow rate, and initial pressure.

Fractional Change Metric for an Aggregate

Following the method of fractional scaling introduced by Zuber et al., (2005), the *aggregate Fractional Rate of Change*, $\overline{\omega}_p$ is obtained by combining all FRCs related to *pressurizing* and *depressurizing* agents of change. The aggregate Fractional Agent of Change $|\overline{\omega}_p|$ gives the combined response rate and scales time correctly for the aggregate of all sub-volumes under the combined action of all change agents.

Depressurizing agents of change are outsurge flow, wall condensation, and heat losses, which cause fluid contraction. *Pressurizing* agent of change is only the heating causing fluid expansion. The INAC 2007, Santos, SP, Brazil.

effects on the vapor and liquid volumes of phenomena like rainout, flashing, and mass transfer through the vapor-liquid interface cancel each other. Thus, by summing over surge outflow and the remaining active agents of change (heat losses, wall condensation, and heating) one obtains the aggregate's Effective Fractional Rate of Change, $\overline{\omega}_p$, for the *response rate of the aggregate*

$$\overline{\omega}_{p} = \omega_{surge} + \omega_{\dot{Q}_{heat}} + \sum_{j=1}^{10} \omega_{j,0}$$
(22)

The Fractional Change Metric of the aggregate for a scaled time phase is then defined as $\Omega_p = |\overline{\omega}_p| t$ and is the measure of fractional change of a state variable (pressure) during the time phase, to which $\overline{\omega}_p$ applies.

The scaled time for the aggregate depressurization and system-level data synthesis is

$$t^* = \Omega_p = \left| \overline{\omega}_p \right| t, \qquad \left| \overline{\omega}_p \right| > 0 \tag{23}$$

According to Eq. (23) the scaled time t^* is stretched in rapid pressure variations and compressed in slow ones, always in accordance with the net dilatation of the fluid caused by *all active agents of change* combined. This is the reason for large pressurizers, like that of IRIS reactor, and small test facilities, like IRIS-PTF, to have a common <u>scaled</u> depressurization history.

For the special case of a surge transient starting from equilibrium conditions, the summation term in Eq. (22) is zero. As a consequence, $\overline{\omega}_p = \omega_{surge} + \omega_{\underline{0}_{heat}} \approx \omega_{surge}$, because surge flow dominates.

The division of Eq. (20) by $\left|\overline{\omega}_{p}\right| = \left|\sum_{j=0}^{11} \omega_{j}\right|_{t=0}$ results:

$$\frac{d p^*}{dt^*} = \sum_{j=0}^{11} \frac{\dot{\mathbf{V}}_j^*}{K_{s,sys}^*} = \sum_{j=0}^{11} \varphi_j^*, \quad p^*(0) = 1$$
(24)

Equation (24) is self-scaling because it has <u>no free scaling parameter</u>. The Fractional Change Metric $\Omega_p = \left|\overline{\omega}_p\right| t$ denotes the *fractional change of pressure* by *all agents of change*. The scaled variable of the *j*th agent of change, φ_i^* , in Eq. (24) is defined as

$$\varphi_{j}^{*} = \frac{\dot{\mathbf{V}}_{j}^{*}}{K_{s,sys}^{*}}$$
(25)

The hierarchical order of the variables φ_j^* is used for ranking agents of change according to their importance, while Eq. (24) has the correct time scaling for data synthesis and possibly for simulating outsurge pressure history.

The condition $|\overline{\omega}_p| > 0$ needed both for Eq. (23) and Eq. (24) will always be met in scaling transients with a direct initiating change agent on a system that starts from steady-state conditions.

4. FRACTIONAL SCALING: COMPONENT LEVEL

At the component level, it is important that the liquid volume is also scaled properly in an experimental facility. The time derivative of the liquid volume is related to the pressure derivative as:

$$\frac{dV_l}{dt} = v_l \frac{dm_l}{dt} + m_l \left(\frac{\partial v_l}{\partial h_l}\right)_p \frac{dh_l}{dt} + m_l \left(\frac{\partial v_l}{\partial p}\right)_{h_l} \frac{dp}{dt}$$
(26)

The substitution of the mass and energy conservation equations in the liquid and Eq. (4) (with i=l) into Eq. (26) results

$$\frac{dV_l}{dt} = B_l - \left(\kappa_{s,l} V_l\right) \frac{dp}{dt}$$
(27)

where

$$B_{l} = \left[\mathbf{v}_{l} - \left(\frac{\partial \mathbf{v}_{l}}{\partial h_{l}}\right)_{p} h_{l} \right] W_{l} + \left(\frac{\partial \mathbf{v}_{l}}{\partial h_{l}}\right)_{p} \dot{Q}_{l}$$
(28)

In view of Eqs. (6) through (17), Eq. (2) and Eq. (28), Eq. (27) can be written as

$$\frac{dV_l}{dt} = \sum_{j=0}^{5} \dot{\mathbf{V}}_j + \dot{\mathbf{V}}_{11} - \frac{\kappa_{s,l} V_l}{V K_{s,sys}} \sum_{j=0}^{11} \dot{\mathbf{V}}_j$$
(29)

The liquid volume is a component state variable (effort). The normalized liquid volume, or *fractional change of liquid volume*, of order of unity, is defined as

$$0 \le V_l^* = \frac{V_l(t) - V_{l\min}}{V_{l\max} - V_{l\min}} = \frac{V_l(t) - V_{l\min}}{\Delta V_{l\max}} \le 1$$
(30)

where $V_{l, \max}$, $V_{l, \min}$ are the maximum and minimum liquid volumes, respectively.

One can obtain the volume scaled equation directly by normalizing each dilatation rate with the *effective dilatation rate of the aggregate*, i.e., the magnitudes of the algebraic sum of all (active) dilatation rates (see Eq. (19)).

Equation (19) is introduced into Eq. (29) that after division by $\Delta V_{l, \text{max}}$ becomes *directly* :

$$\frac{dV_l^*}{dt} = \frac{\left|\sum_{j=0}^{11} \dot{\mathbf{V}}_j\right|_{t=0} \left(\sum_{j=0}^5 \dot{\mathbf{V}}_j^* + \dot{\mathbf{V}}_{11}^*\right)}{\Delta V_{l,\max}} - \frac{\left|\sum_{j=0}^{11} \dot{\mathbf{V}}_j\right|_{t=0} \kappa_{s,l,0} \kappa_{s,l}^* V_l^*}{V K_{s,sys,0} \kappa_{s,sys}^*} \sum_{j=0}^{11} \dot{\mathbf{V}}_j^*$$
(31)

Eq. (31 can be written as

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$$\frac{dV_l^*}{dt} = \left|\overline{\omega}_V\right| \times \left(\sum_{j=0}^5 \varphi_j^{*V} + \varphi_{11}^{*V} - \sum_{j=0}^{11} \varphi_j^{*p}\right)$$
(32)

The scaled variable of the j^{th} agents of change, φ_j^{*v} and φ_j^{*p} , in Eq. (32) <u>depends on the FRCs of all</u> <u>actives agents of change</u> as:

$$\varphi_j^{*V} = \frac{\omega_{j,0}^V}{\left|\overline{\omega}_V\right|} V_j^* \qquad (33) \text{ and } \qquad \varphi_j^{*p} = \frac{\omega_{j,0}^p}{\left|\overline{\omega}_V\right|} \frac{\kappa_{s,l}^* V_l^*}{K_{s,sys}^*} \dot{V}_j^* \qquad (34)$$

The FRCs of all actives agents of change are:

$$\omega_{j,0}^{v} = \begin{cases} \frac{\dot{V}_{j,0}}{\Delta V_{l,\max}}, \ j = 0, \dots, 5, \text{ and } j = 11 \\ 0, \text{ otherwise} \end{cases}$$
(35)
$$\omega_{j,0}^{p} = \frac{\kappa_{s,l,0} \dot{V}_{j,0}}{V K_{s,sys,0}}, \ j = 0, \dots, 11$$
(36)

The Aggregate Fractional Rate of Change $\overline{\omega}_{V}$ is defined below.

4.1. Aggregate Fractional Change Metric for Liquid Volume

Following the method of fractional scaling introduced by Zuber et al., (2005), the *aggregate Fractional Rate of Change*, $\overline{\omega}_{V}$ is obtained by adding all positive FRCs related to *pressurizing* agents of change to the sum all negative FRCs related to *depressurizing* agents of change. The aggregate Fractional Agent of Change $|\overline{\omega}_{V}|$ gives the correct aggregate response rate and scales time correctly for the aggregate of all "*processes*" under the action of all agents of change combined. It includes the action of all agents of change, which are active during the scaled time phase.

Contracting agents of change are outsurge flow, wall condensation, and heat losses, which cause fluid contraction. *Expanding* agent of change is heating which causes overall fluid expansion. The effects on liquid volume of phenomena like rainout, flashing, and mass transfer through the vapor-liquid interface cancel each other. Thus, by summing over surge outflow and the remaining active agents of change (heat losses, wall condensation, and heating), or by scaling according to Eq. (31), one obtains the aggregate's Effective Fractional Rate of Change, $\overline{\omega}_V$, for the *response rate of the aggregate* processes

$$\overline{\omega}_{V} = \omega_{surge}^{V} + \omega_{\dot{Q}_{heat}}^{V} + \sum_{j=1}^{5} \omega_{j,0}^{V} - \omega_{surge}^{p} - \omega_{\dot{Q}_{heat}}^{p} - \sum_{j=1}^{10} \omega_{j,0}^{p} = \omega_{surge}^{V} + \omega_{\dot{Q}_{heat}}^{V} + \sum_{j=1}^{10} \omega_{j,0}^{V}$$
(37)

where

$$\boldsymbol{\omega}_{j}^{V} = \boldsymbol{\omega}_{j}^{V} - \boldsymbol{\omega}_{j}^{p}$$
(38)

as can be seen from Eqs. (35) and (36).

The Fractional Change Metric of the aggregate processes for a scaled time phase is then defined as $\Omega_V = |\overline{\omega}_V| t$ and is the measure of fractional change of a state variable (liquid volume) during the INAC 2007, Santos, SP, Brazil.

time phase, to which $\overline{\omega}_v$ applies. The Fractional Change Metric is negative during depressurization and positive for a pressure excursion when net heating (heating minus wall condensation and heat losses) overwhelms outsurge flow. If the Effective Fractional Rate of Change $\overline{\omega}_v$ is close to zero for a time phase, then the Fractional Change Metric also approaches $\Omega_v \approx 0$, and the associated time phase has nearly constant liquid volume. This occurs in the time phase when the surge flow returns to zero, after pressure control is achieved and the heating is just that necessary to compensate for wall condensation and heat losses through the walls of pressurizer.

The scaled time for the aggregate processes on liquid volume and component-level data synthesis is

$$t^* = \Omega_V = \left| \overline{\omega}_V \right| t, \qquad \left| \overline{\omega}_V \right| > 0 \tag{39}$$

where the time counts from the beginning of the time phase. According to Eq. (39) the scaled time t^* is stretched in rapid and compressed in slow liquid volume changes, always in accordance with the net dilatation of the fluid caused by *all active agents of change* combined. This is the reason for large pressurizers, like that of IRIS reactor, and small test facilities, like IRIS-PTF, to have a common scaled liquid volume history.

Division of Eq. (32) by $|\overline{\omega}_{V}|$ and using Eq. (39) gives

$$\frac{dV_l^*}{dt^*} = \sum_{j=0}^5 \varphi_j^{*V} + \varphi_{11}^{*V} - \sum_{j=0}^{11} \varphi_j^{*p}, \quad V_l^*(0) = 1$$
(40)

Equation (40) is self-scaling because it has <u>no free scaling parameter</u>. The Fractional Change Metric $\Omega_v = |\overline{\omega}_v| t$ denotes the *fractional change of liquid volume* by *all agents of change*. Time t^* in Eq. (40) is stretched or compressed according to the net dilatation caused by all agents of change active during a time phase, as explained below Eq. (39).

Equation (40) has the correct time scaling for data synthesis and possibly for simulating outsurge liquid volume history. Again the condition $|\overline{\omega}_V| > 0$ needed both for Eq. (39) and Eq. (40) will be met in scaling transients with a direct initiating change agent on a system that starts from steady-state conditions. To be similar to IRIS pressurizer within the FSA framework, a scaled experiment must have the same dominants FRCs of surge flow and heating as that of IRIS pressurizer. From Eqs. (6) and (21), the FRC of surge flow is defined as:

$$\boldsymbol{\omega}_{surge} = \frac{\left(\mathbf{v}_{l} - \left(\frac{\partial \mathbf{v}_{l}}{\partial h_{l}}\right)_{p} \left(h_{l} - h_{surge}\right)\right) \dot{m}_{surge, \max}}{VK_{s, sys, 0} \Delta p}$$
(41)

From Eqs. (17) and (21), the FRC of heating is defined as

$$\omega_{heat} = \frac{\left(\frac{\partial \mathbf{v}_l}{\partial h_l}\right)_p (\dot{Q}_{heat, \max})}{VK_{s, sys, 0} \Delta p}$$
(42)

As the experiment has the same pressure, same thermodynamic properties, same system elasticity and same pressure range as IRIS pressurizer, it results from Eqs. (41) and (42) that for similarity it is necessary that the surge mass flow rate and the heating power keep the same proportion of the volume ratio.

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5. APPLICATION

FSA at the system level is applied here in two ways, namely for establishing a *process hierarchy* and *assessing scale distortions* on the basis of Fractional Rate of Change (FRC) and for *data synthesis* on the basis of the Fractional Change Metric. The case analyzed was the IRIS pressurizer (IRIS-P) and IRIS-PTF, an experimental test facility that is being designed to test IRIS pressurizer performance and safety, whose volume is 1/100 of that of IRIS pressurizer. The basic transient studied was a typical outsurge transient in IRIS pressurizer (Barroso and Batista, 2004).

The FRC-based hierarchy of agents of change, or transfer processes across system boundaries is an *objectively quantified* alternative and a step forward, from the expert opinion based Phenomena Identification and Ranking Table (PIRT). Moreover, the FRCs serves to *identify* and *quantify scale distortion*. The graph $p^*(|\Omega_p|) = p^*(t^*_{surge})$ should reveal whether the fractional change of pressure p^* during outsurge tests is the same at the same value of the Fractional Change Metrics of outsurge, Ω_p , regardless of IRIS pressurizer and test facility size and actual outsurge time. By the same token, the graph $V_i^*(|\Omega_v|) = V_i^*(t^*_{surge})$ should reveal whether the fractional change of liquid volume V_i^* is the same at the same value of the Fractional Change Metrics of outsurge, Ω_v . Congruence of respective curves $p^*(|\Omega_p|)$ and $V_i^*(|\Omega_v|)$ indicates that the (out)-surge flow governs the entire transient and that system volume, system elasticity, (dominated by saturated liquid volume in pressurizer) maximum outsurge flow, maximum heating power, initial liquid volume, and initial pressure are properly scaled; differences reveal scale distortions. The initial slopes of $p^*(|\Omega_p|)$ and $V_i^*(|\Omega_v|)$ should be always (-1).

5.1 Initial Conditions of Pressurizer

Table 1 presents estimates of mass flow rates due to local phenomena of wall condensation, flashing, and rainout in IRIS-P and IRIS-PTF at steady state condition.

	W _{WC}	$W_{ m FL}$	W _{RO}
	(10^{-3} kg/s)	(10^{-3} kg/s)	(10^{-3} kg/s)
IRIS-P	1.4189	9.7385	8.4217
IRIS-PTF	0.096667	0.48658	0.39096

Table 1. Mass Flow Rates in IRIS Pressurizer

Due to the equality of thermal constants, the bubble terminal velocity in the liquid volume, if calculated using the Zuber formula (*in* Whalley, 1996) would be the same in IRIS-P and IRIS-PTF. The interface area ratio in IRIS-P and in IRIS-PTF is 21.5. In a calculation, with same bubble terminal velocity, the flashing mass flow rate ratio in IRIS-P and in IRIS-PTF would also be 21.5. Based on Eq. (15), the flashing mass flow rate ratio in IRIS-P and in IRIS-PTF is (9.7385/0.4866) = 20 (see Table 1), which is very close to the value that would be obtained using the Zuber formula. From Table 1, the rainout mass flow rate in IRIS-P divided by that in IRIS-PTF is $(8.4217/0.3901) \approx 21.6$, which is very close to the interface area ratio in IRIS-PTF. Table 2 exhibits the heater power necessary to compensate wall condensation and wall losses, interface heat, and interface mass flow rate, at steady state conditions, and the maximum interface mass flow rate

	$\dot{Q}_{heat,0}$	\dot{Q}_{lv}	\dot{m}_{lv}
	(kW)	(W)	(kg/s)
IRIS-P	28.710	0.00	4.83E-9
IRIS-PTF	1.3623	0.00	4.83E-10

Table 2. Heater Power and Interface Flow at Steady State

Table 3 exhibits the maximum surge mass flow rate in a out-surge transient and the maximum heating provided by the heater control system.

	$\dot{Q}_{heat,\mathrm{max}}$	$\dot{m}_{surge,\mathrm{max}}$
	(kW)	(kg/s)
IRIS-P	1000	-54.839
IRIS-PTF1	13.333	-0.73118
IRIS-PTF2	10.0	-0.54839

 Table 3. Maximum Surge Flow and Maximum Heater Power

The out-surge transient in IRIS-PTF1 has a maximum mass flow rate and maximum heater power that are 1/75 of that of IRIS-P. The maximum surge flow rate of an out-surge transient in the experiment is such that the dominant FRC of surge flow given by Eq. (40) in IRIS-PTF2 is equal to that in IRIS-P. The maximum heating power in the experiment is such that the FRC for heating given by Eq. (41) in IRIS-PTF2 be equal to that in IRIS-P. Therefore, the maximum surge flow ratio and the maximum heating power in IRIS-PTF2 keep the same volume ratio of 1/100 in relation to IRIS-P.

5.2 Establishing Process Hierarchy and Assessing Scale Distortion

The outsurge flow leaving the system is the dominant agent of change. Table 4 lists associated numerical values of the maximum and aggregate *fractional rate of change*, ω_{surge} , Eq. (41), and $\overline{\omega}_{p}$, Eq. (22). Also shown in Table 4 are the residence time, or turn-over time, $\Delta t_{surge} = V/\dot{V}_{surge}$, and the fractional volume compression, $\Delta p K_{s,sys,0}$. At the surge volume rate, \dot{V}_{surge} , to displace the control volume *V*, it takes one residence time.

	Resid. Time	$\Delta p K_{s, sys, 0}$	ω_{surge}	$\overline{\mathcal{O}}_p$
	Δt_{surge} (s)	-	$(10^{-3} \%/s)$	$(10^{-3} \%/s)$
IRIS-P	851.57	0.12054	-9.7422	-8.8548
IRIS-PTF1	638.68	0.12338	-12.690	-11.534
IRIS-PTF2	851.57	0.12338	-9.5175	-8.6506

Table 4. Fractional Rates of Change for Surge Flow

It can be seen in Table 4 that the residence time dominates the inverse of ω_{surge} , the FRC of maximum surge flow in both facilities of different sizes, when the same fluid is used at the same pressure. The fractional volume compression $\Delta p K_{s,sys,0}$ is the fractional reduction of fluid volume when it is adiabatic compressed by Δp .

From Eq. (41) it can be demonstrated that the inverse of ω_{surge} is the product of the fractional volume compression times the residence time, or $(1/\omega_{surge}) = (\Delta p K_{s, sys, 0}) \Delta t_{surge}$. This relationship can be readily verified using the data in Table 4. As the fractional volume compression $\Delta p K_{s, sys, 0}$ is small, this explains why the residence time dominates the inverse of ω_{surge} .

Tables 5 and 6 exhibit the FRCs of pressure in Eq. (20) for the three facilities (IRIS-P, IRIS-PTF1 and IRIS-PTF2) that are active (although their sum is zero) at the start of the out-surge transient, together with the FRC of maximum surge flow and maximum heat input that are reached some time after the start of the transient. The **surge flow** leads and is followed by **heat input** provided by the pressurizer heater control system.

	Surge Flow	Wall Condensation	Vapor Rainout	Liquid Flash	Heat Loss	Interface Flow
	ω_{surge}	$\omega_{l,WC}$	$\omega_{l,RO}$	$\omega_{l,FL}$	ω_{l,\dot{Q}_l}	$\omega_{l, lv}$
	10^{-3} /s	10^{-3} /s	10^{-3} /s	10^{-3} /s	10^{-3} /s	10^{-3} /s
Index j	0	1	2	3	4	5
IRIS-P	-9.7422	0.00025	0.00149	010075	.008434	8.6E-12
-PTF1	-12.690	0.00168	0.00678	049178	0.04082	8.39E-11
-PTF2	-9.5175	0.00168	0.00678	049178	0.04082	8.39E-11

Table 5. Fractional Rates of Change of Pressure

Table 6.	Fractional	Rates of	Change	of Pressure
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	Liquid	Wall	Vapor	Heat	Interface	Heat
	Flash	Condensation	Rainout	Loss	Flow	Input
	$\mathcal{O}_{\mathrm{v},FL}$	$\mathcal{O}_{\mathrm{v},WC}$	$\mathcal{O}_{\mathrm{v},RO}$	$\omega_{v,\dot{Q}_{v}}$	$\mathcal{O}_{\mathrm{v},l\mathrm{v}}$	$\omega_{\dot{Q}_{heat}}$
	10^{-3} /s	10^{-3} /s	10^{-3} /s	10^{-3} /s	10^{-3} /s	10^{-3} /s
Index j	6	8	7	9	10	11
IRIS-P	0.01007	-0.00147	001496	007219	-8.6E-12	0.88735
-PTF1	0.04918	-0.00977	006785	032733	-8.4E-11	1.1558
-PTF2	0.04918	-0.00977	006785	032733	-8.4E-11	0.86688

Table 7 exhibits the $\hat{\omega}_j = \omega_j / \omega_{surge}$, the ratio of some FRCs to the surge flow (maximum) FRC, ω_{surge} . The ratios $|\hat{\omega}_j|$ can be ordered in a hierarchy. A criterion, such as $|\hat{\omega}_j| > 0.1$, is used to establish that only the **surge out-flow** and **input heating** are **important** agents of change, the others are **unimportant**.

	Wall	Vapor	Liquid	Heat	Interface	Heat
	Condensation	Rainout	Flash	Loss	Flow	Input
	$\hat{\pmb{\omega}}_{l,WC}$	$\hat{\pmb{\omega}}_{\!\scriptscriptstyle l,RO}$	$\hat{\pmb{\omega}}_{l,FL}$	$\hat{\omega}_{l,\dot{Q}_{l}}$	$\hat{\pmb{\omega}}_{l,l\mathrm{v}}$	$\hat{\pmb{\omega}}_{\dot{Q}_{heat}}$
Index j	1	2	3	4	5	11
IRIS-P	.000026	.001536	010341	.000866	8.8E-13	0.09108
-PTF1	.000132	.005347	038753	.003217	6.6E-12	0.09108
-PTF2	0.00018	.007129	051671	.004289	8.8E-12	0.09108

Table 7. Fractional Rates of Change Ratios for Pressure

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Tables 8 and 9 exhibit the FRCs of liquid volume) in Eqs. (35) and (36) for the three facilities (IRIS-P, IRIS-PTF1 and IRIS-PTF2 that are active (although their sum is zero) at the start of the out-surge transient, together with the FRC of maximum surge flow and maximum heat input that are reached some time after the start of the transient.

	Surge Flow	Wall Condensation	Vapor Rainout	Liquid Flash	Heat Loss	Interface Flow
	ω_{surge}^{V}	$\omega_{\scriptscriptstyle l,WC}^{\scriptscriptstyle V}$	$\omega_{l,RO}^{V}$	$\omega_{l,FL}^{V}$	ω_{l,\dot{Q}_l}^{V}	$\omega_{l, lv}^{V}$
	10^{-3} /s	10^{-3} /s	10^{-3} /s	10^{-3} /s	10^{-3} /s	10^{-3} /s
Index j	0	1	2	3	4	5
IRIS-P	-1.3854	.000036	0.00021	-0.00143	0.00120	1.2E-12
-PTF1	-1.8256	0.00024	0.00098	-0.00707	0.00587	1.21E-11
-PTF2	-1.3692	0.00024	0.00098	-0.00707	0.00587	1.21E-11

Table 8. Fractional Rates of Change of Liquid Volume

Table 9. Fractional Rates of Change of Liquid Volume

	Liquid	Wall	Vapor	Heat	Interface	Heat
	Flash	Condensation	Rainout	Loss	Flow	Input
	$\pmb{\omega}_{\mathrm{v},\mathit{FL}}^{\!\scriptscriptstyle V}$	$\boldsymbol{\varTheta}_{\mathrm{v},WC}^{V}$	$\boldsymbol{\omega}_{\mathrm{v},RO}^{V}$	$\omega^{\scriptscriptstyle V}_{\scriptscriptstyle { m v},\dot{\mathcal{Q}}_{\scriptscriptstyle m v}}$	$\omega_{\mathrm{v},l\mathrm{v}}^{\mathrm{v}}$	$\omega^{\scriptscriptstyle V}_{{\dot Q}_{heat}}$
	10^{-3} /s	10^{-3} /s	10^{-3} /s	10^{-3} /s	10^{-3} /s	10^{-3} /s
Index j	6	8	7	9	10	11
IRIS-P	002753	.000401	.000409	.001973	2.3E-12	0.12619
-PTF1	013595	0.00270	.001876	.009049	2.32E-11	0.16628
-PTF2	013595	0.00270	.001876	.009049	2.32E-11	0.12471

The criterion for scale distortion cannot be tighter than is permitted by the compound of pertinent experimental uncertainties, manufacturing tolerances and operational uncertainties, or else no important process can ever be found to be without scale distortion. The firs two rows of results in Tables 5 and 6 show that the **out-surge depressurization** processes **of importance** in IRIS-PTF1 are distorted, because the difference between the corresponding **important** FRCs of **surge out-flow** (30%) and **input heating** (30%) are both greater than 10%. Nevertheless, in IRIS-PTF2, the first and the last rows in Tables 5 and 6 show **no out-surge depressurization** processes **of important** FRCs, **surge out-flow** (2.36%) and **input heating** (2.36%) are both less than 10%.

From the agents of change for liquid volume in Tables 8 and 9 we see again that the **surge flow** leads and is followed by **heat input** provided by the pressurizer heater control system.

The first two rows of results in Tables 8 and 9 show that the **out-surge depressurization** processes **of importance** are distorted in IRIS-PTF1, because the difference between the two corresponding **important** FRCs, **surge out-flow** (32%) and **input heating** (32%) are both greater than 10%. Nevertheless, in IRIS-PTF2, the first and the last rows of results in Tables 8 and 9 show **no out-surge depressurization** processes **of importance** that is distorted, because the difference between the two corresponding **important** FRCs, **surge out-flow** (1.18%) and **input heating** (1.19%) are both less than 10%.

5.3 Data Synthesis

Figures 2 and 3 show, in currently prevailing and familiar format, calculated typical out-surge pressure histories, i.e., pressure versus out-surge time, for IRIS pressurizer (IRIS-P) and 1/100 volume scaled pressure test facilities (IRIS-PTF1 and IRIS-PTF2). The corresponding liquid volume histories, i.e., liquid volumes versus out-surge time are shown in Figures 4 and 5.



Figure 2. Calculated IRIS-P and –PTF1 Depressurizations: scaled facility effects do not preserve out-surge flow time.



Figure 3. Calculated IRIS-P and -PTF2 Depressurizations: facility effects are scaled to preserve out-surge flow time.



Figure 4. Calculated IRIS-P and –PTF1 Liquid Volume Variations: scaled facility effects do not preserve time of liquid volume variation.



Figure 5. Calculated IRIS-P and –PTF2 Liquid Volume Variations: facility effects are scaled to preserve time of liquid volume variation.

Figures 2 and 3 present the effects of facility sizes on pressure history. Scaled IRIS-PTF1 surge flow level and pressure control system did not preserve *time for out-surge transient*, because $|\omega_{surge}|$

is 0.0097422/s for IRIS-P and 0.012690/s for IRIS-PTF1 (Table 5), which is 30% more, and thus outside the accepted uncertainty range of surge flow determination (>10%).

Figures 4 and 5 present the effects of facility sizes on liquid volume histories. Again, time scaling is not preserved in Figure 4 for IRIS-PTF1 because $|\omega_{surge}^{v}|$ is 0.0013854 in IRIS-P and 0.0018256 in IRIS-PTF1, but is nearly preserved in IRIS-PTF2, because now $|\omega_{surge}^{v}|$ is 0.0013692 (Table 8). As can be seen in Figure 5, the initial 1/100 volume ratio in IRIS-PTF2 is almost preserved. Evidently, scaled IRIS-PTF2 surge flow level and pressure control system nearly preserved pressure and *time*

for out-surge transient, because $|\omega_{surge}|$ is 0.0095175/s for IRIS-PTF2 (Table 5), which is 2.36% less, close and within the accepted uncertainty range of surge flow determination.

The Fractional Rates of Change (FRC) in Tables 5 and 6 were used to calculate the FRC of the aggregate, $\overline{\omega}_p$ in Table 4, to form the Fractional Change Metric $\Omega_p = |\overline{\omega}_p| t$ and scale the time of the calculated transient in Figures 2 and 3 according to Eq. (24). The pressure was scaled according to Eq. (35). This leads to the plots of $p^*(\Omega_p)$ in Figures 6 and 7, which shows the effects of scaling parameters $\hat{\omega}_i$.







Figure 7. Scaled Out-surges in IRIS-P and IRIS-PTF2: effects of surge mass flow rate on time of out-surge flow are scaled.

5.4 Scaling of Surge Mass Flow Rate

For IRIS-PTF the effect of out-surge mass flow rate (also of heater power) on pressure is scaled, because the two curves in Figures 6 and 7 almost collapse into one, even though the times of out-surge differ by 25% in Figure 2. For IRIS-PTF the effect of out-surge mass flow rate and of heater power on liquid volume is also scaled, because the two curves in Figures 8 and 9 almost collapse into one, even though the times of out-surge differ by 25% in Figure 4.

Nevertheless, the more pronounced shape differences between the IRIS-PTF1 curves in Figures 2 and 4 suggest *dissimilar* processes during out-surges in which the mass flow rates (and heater power) are not scaled according to the pressurizer volume ratio of 1/100, as in IRIS-PTF2. The convergence between the IRI-PTF2 and IRIS-P curves in Figures 3 and 5 is a consequence of the almost agreement between the important FRCs in IRIS-P and IRIS-PTF2. Thus, for pressure and liquid level effects, IRIS-PF2 is *similar* to the IRIS pressurizer.

The Fractional Rates of Change (FRC) in Tables 7 and 8 were used to form the Fractional Change Metric $\Omega_v = \overline{\omega}_v t$ and scale the time of the calculated transient in Figures 4 and 5 according to Eq. (40). This leads to the plots of $V_l^*(\Omega_v)$ in Figure 8 and 9, which shows the effects of scaling parameters $\hat{\omega}_i^v$.

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Figure 8. Scaled Out-surges in IRIS-P and IRIS-PTF1: effects of liquid volume on time of liquid volume variation are scaled.



Figure 9. Scaled Out-surges in IRIS-P and IRIS-PTF2: effects of liquid volume on time of liquid volume variation are scaled.

6. CONCLUSIONS

We have applied Fractional Scaling Analysis (FSA) at the system-level, first on the basis of Fractional Rates of Change ω_j (FRC) to identify, quantify and prioritize in the order of their impact on all transfer processes responsible for system depressurization (see Tables 5, 6 and 7), and secondly on the basis of the Fractional Change Metric Ω_p to synthesize computer calculated data (compare Figures 2 and 3 with Figures 6 and 7). We have demonstrated quantitatively for a typical out-surge in IRIS pressurizer and a test installation

being designed, called IRIS-PTF of 1/100 volume, that the surge flow is the leading agent of change for out-surge depressurization. It turned out that the absolute value of the out-surge flow-related Fractional Change Metric $\Omega_{surge} = t_{surge}^*$, alone synthesizes the pressure history.

We have used <u>local scaling</u> and acknowledge the role that it plays in the calculation of the heating power of the experiment, once the size and the surge flow data of the experiment are defined.

We also have applied Fractional Scaling Analysis (FSA) at the component-level, first on the basis of Fractional Rates of Change ω_j^V (FRC) to identify, quantify and prioritize in the order of their impact on all transfer processes responsible for liquid volume variation (see Tables 8 and 9), and secondly on the basis of the Fractional Change Metric Ω_V to synthesize computer calculated data (compare Figures 4 and 5 with Figures 8 and 9). We have demonstrated quantitatively for a typical out-surge in IRIS pressurizer and the test installation IRIS-PTF, that the surge flow is also the leading agent of change for liquid volume variation. It turned out that the absolute value of the out-surge flow-related Fractional Change Metric $\Omega_{surge}^V = t_{surge}^{*V}$, alone synthesizes the liquid volume history.

We have found that a successful data synthesis requires that the surge flow (and heating power) be scaled in the same proportion to the pressurizer and test installation volume ratio. A successful data synthesis also requires a reliable and complete system specification and test data and is therefore performed most efficiently and reliably during testing, when data can be easily retrieved as part of a computer simulation.

NOMENCLATURE

h enthalpy per unit mass $D_{I_{i}, max}$ ratig on infinite rotations h_{I_v} interface enthalpy per unit mass Φ_j rate of pressure change effect from j^{th} agent of change $h_{Nusselr}$ heat transfer coefficient of wall condensation Φ_j rate of pressure change scaled to include fixed ratio of fractional change rates $M_{Nusselr}$ interface mass flow rate φ_j^* j^{th} rate of pressure change scaled to include fixed ratio of fractional change rates \dot{m}_{iv} interface mass flow rate K_v specific isentropic compressibility of liquid \dot{m}_{surge} surge mass flow rate K_v specific isentropic compressibility of vapor \dot{p} pressure K_s , specific isentropic compressibility of system \dot{Q}_{heater} heating power Ω_p Fractional Change Metric of pressure state variable \dot{Q}_{iv} interface heat transfer power Ω_p Fractional Change Metric liquid volume \dot{Q}_{vv} heat balance in vapor volume $\overline{\omega}_p$ effective rate of fractional change \dot{Q}_{vv} wall heat losses from liquid volume $\overline{\omega}_p$ effective rate of fractional change of liquid volume \dot{Q}_{vv} wall heat losses from vapor volume $\hat{\omega}_j$ j^{th} ratio of rates of fractional change \dot{S}_v vapor volume wall area $\hat{\omega}_j$ j^{th} ratio of rates of fractional change J_v inquid volume wall area $\hat{\omega}_j$ j^{th} ratio of rates of fractional change	h h_{lv}	enthalny per unit mass	$\Delta v_{l, \max}$ range of inquite volumes
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surgeKSurgeK p pressure Q_p Fractional Change Metric of pressure state \dot{Q}_{heater} heat balance in liquid volume Ω_p Fractional Change Metric liquid volume \dot{Q}_l heat balance in liquid volume Ω_V Fractional Change Metric liquid volume \dot{Q}_{lv} interface heat transfer power ω_j j^{th} rate of fractional change \dot{Q}_v heat balance in vapor volume $\overline{\omega}_p$ effective rate of fractional change of pressure \dot{Q}_{wl} wall heat losses from liquid volume $\overline{\omega}_V$ effective rate of fractional change of liquid \dot{Q}_{wv} wall heat losses from vapor volume $\overline{\omega}_j$ j^{th} ratio of rates of fractional change S_v vapor volume wall area $\hat{\omega}_j$ j^{th} ratio of rates of fractional change S_v vapor volume wall area $\hat{\omega}_j$ j^{th} ratio of rates of fractional change T fluid temperatureSubscripts	\dot{m}_{surage}	surge mass flow rate	K_{v} specific isentropic compressibility of vapor
$ \begin{array}{cccc} \dot{Q}_{heater} & \mbox{heating power} & & \Omega_p & \mbox{Fractional Change Metric of pressure state} \\ \dot{Q}_l & \mbox{heat balance in liquid volume} & & \Omega_V & \mbox{Fractional Change Metric liquid volume} \\ \dot{Q}_{lv} & \mbox{interface heat transfer power} & & & & & & \\ \dot{Q}_v & \mbox{heat balance in vapor volume} & & & & & \\ \dot{Q}_v & \mbox{heat balance in vapor volume} & & & & \\ \dot{Q}_{wl} & \mbox{wall heat losses from liquid volume} & & & & \\ \dot{Q}_{wv} & \mbox{wall heat losses from vapor volume} & & & \\ \dot{Q}_{wv} & \mbox{wall heat losses from vapor volume} & & & \\ \dot{Q}_v & \mbox{effective rate of fractional change of pressure} \\ \dot{Q}_w & \mbox{wall heat losses from vapor volume} & & \\ \dot{Q}_v & \mbox{volume} & & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} \\ \dot{Q}_v & \mbox{volume} & & \\ \dot{Q}_v & \mbox{volume} & & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{volume} & & \\ \dot{Q}_v & \mbox{volume} & & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{volume} & & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} & \\ \dot{Q}_v & \mbox{effective rate of fractional change} &$	p	pressure	$\mathbf{K}_{s, syst}$ specific isentropic compressibility of system
\dot{Q}_l heat balance in liquid volumevariable \dot{Q}_l interface heat transfer power Ω_V Fractional Change Metric liquid volume \dot{Q}_{lv} heat balance in vapor volume ω_j j^{th} rate of fractional change \dot{Q}_v heat balance in vapor volume $\overline{\omega}_p$ effective rate of fractional change of pressure \dot{Q}_{wl} wall heat losses from liquid volume $\overline{\omega}_p$ effective rate of fractional change of liquid \dot{Q}_{wv} wall heat losses from vapor volumevolume \dot{Q}_{uv} vall heat losses from vapor volume $\overline{\omega}_j$ \dot{S}_l liquid volume wall area $\hat{\omega}_j$ j^{th} ratio of rates of fractional change S_v vapor volume wall area $\hat{\omega}_j$ j^{th} ratio of rates of fractional change T fluid temperatureSubscripts	\dot{Q}_{heater}	heating power	Ω_p Fractional Change Metric of pressure state
$\begin{array}{cccc} \dot{\mathcal{L}}_{l} & & & & & & \\ \dot{\mathcal{L}}_{l} & & & & & \\ \dot{\mathcal{L}}_{l} & & \\ \dot{\mathcal{L}}_{l} & & & \\ $	\dot{O}_i	heat balance in liquid volume	variable
$\begin{array}{cccc} \dot{\mathcal{Q}}_{lv} & \text{harthace heat function power} & & & & & & & & & & & & & & & & & & &$	\dot{o}^{l}	interface heat transfer nower	Ω_V Fractional Change Metric liquid volume
$\begin{array}{c} Q_{\rm v} & \text{heat balance in vapor volume} & \overline{\omega}_{p} & \text{effective rate of fractional change of pressure} \\ \dot{Q}_{wl} & \text{wall heat losses from liquid volume} & \overline{\omega}_{V} & \text{effective rate of fractional change of liquid} \\ \dot{Q}_{wv} & \text{wall heat losses from vapor volume} & \text{volume} \\ S_{l} & \text{liquid volume wall area} & \hat{\omega}_{j} & j^{\text{th}} \text{ ratio of rates of fractional change} \\ S_{\rm v} & \text{vapor volume wall area} & Subscripts \\ T & \text{fluid temperature} \end{array}$	\mathcal{Q}_{lv}		$\omega_j \qquad j^{\rm th}$ rate of fractional change
Q_{wl} wall heat losses from liquid volume $\overline{\omega}_{V}$ effective rate of fractional change of liquid \dot{Q}_{wv} wall heat losses from vapor volumevolume S_{l} liquid volume wall area $\hat{\omega}_{j}$ j^{th} ratio of rates of fractional change S_{v} vapor volume wall area $Subscripts$ T fluid temperatureSubscripts	\mathcal{Q}_{v}	heat balance in vapor volume	$\overline{\boldsymbol{\omega}}_p$ effective rate of fractional change of pressure
\dot{Q}_{wv} wall heat losses from vapor volumevolume S_l liquid volume wall area $\hat{\omega}_j$ j^{th} ratio of rates of fractional change S_v vapor volume wall areaSubscripts T fluid temperatureSubscripts	Q_{wl}	wall heat losses from liquid volume	$\overline{\omega}_{V}$ effective rate of fractional change of liquid
S_l liquid volume wall area $\hat{\omega}_j$ j^{th} ratio of rates of fractional change S_v vapor volume wall areaSubscripts T fluid temperatureSubscripts	\dot{Q}_{wv}	wall heat losses from vapor volume	volume
S_v vapor volume wall areaSubscriptsTfluid temperatureSubscripts	S_l	liquid volume wall area	$\hat{\boldsymbol{\omega}}_{j}$ $j^{ ext{th}}$ ratio of rates of fractional change
T fluid temperature	$S_{\rm v}$	vapor volume wall area	Subscripts
	Т	fluid temperature	Subscripts
t time FL flashing	t	time	FL flashing
v volume per unit mass RO rainout	V	volume per unit mass	RO rainout
V total volume of pressurizer WC wall condensation	V	total volume of pressurizer	WC wall condensation
V liquid volume f saturated liquid	V_{i}	liquid volume	f saturated liquid
g saturated vapor	IZ	1	g saturated vapor
v _v vapor volume heat heater power	V _v	vapor volume	heat heater power
V j th rate of volume change i liquid or vapor	$\dot{\mathbf{V}}_{i}$	j th rate of volume change	i liquid or vapor
j factor of change	, , ,	th an a final fi	j factor of change
V_j j^{av} normalize rate of volume change l inquid	\mathbf{V}_{j}	$j^{\rm m}$ normalize rate of volume change	<i>l</i> liquid
W_l mass balance $t=0$ initial value	W_l	mass balance	t=0 initial value
W_{FL} flashing mass flow rate v vapor	W_{FL}	flashing mass flow rate	v vapor
W_{RO} rainout mass flow rate	W_{RO}	rainout mass flow rate	
W_{WC} wall condensation mass flow rate Superscripts	W_{WC}	wall condensation mass flow rate	Superscripts
*			* cooled
Greek Symbols Scaled	Greek S	Symbols	 scaled non-dimensional variable

Δp	range of pressures
Δt_{ref}	reference time

non-unn	chistofiai	variable	
changes	of liquid	volume	dr

- р v
- changes of liquid volume due to pressure changes of liquid volume due to specific volume changes of liquid volume due to v and p, Eq. (67) V

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