

EFFECT OF MICROWAVE PROCESSING ON THE PROPERTIES OF NICKEL OXIDE/ZIRCONIA/CERIA COMPOSITES

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Introduction

Ni/ZrO₂:8 mol%Y₂O₃ (Ni/YSZ) cermet is the standard anode for solid oxide fuel cells (SOFC). Such composites have excellent properties, unheard by any other alternative material proposed for H₂ fuelled SOFCs. Chemical syntheses are preferred as compared to solid state methods to attain the highly dispersed Ni particles and homogenous phase distribution. The coprecipitation method can provide homogeneous and small-sized powders for SOFC anodes with superior properties. In the present study, NiO and CeO₂ were synthesized by co-precipitation in a suspension of YSZ, forming the NiO/YSZ/CeO₂ composite. The coprecipitation was followed by a microwave-assisted hydrothermal treatment. The effect of the MW was investigated and the results were compared with samples prepared by conventional heat treatment.

Experiments

NiO/YSZ/CeO₂ (60/20/20 wt.%) composites were prepared by coprecipitation. YSZ (Tosoh) powder was kept under stirring in NH₄OH solution in which nickel acetate and cerium nitrate (Aldrich) water solutions were added dropwise. After washing with water and ethanol the as-prepared powder was treated in a conventional oven (CT) or in a microwave-assisted hydrothermal (MW) reactor, both at 130 °C for 2 h. Calcining was carried out at 500 °C for 4 h in air. Uniaxially pressed cylindrical pellets were sintered at 1250 °C for 3 h. Samples were studied by simultaneous thermogravimetric and differential thermal analysis (TG/DTA) and X-Ray diffractometry. The linear retraction during sintering of green ceramic compacts was followed by dilatometry up to 1400 °C. The electrical properties of sintered samples were studied by impedance spectroscopy in the 200-700 °C temperature range.

Results and Discussion

Figure 1 shows the TG/DTA data in the temperature range in which the main mass loss (~10 %) of coprecipitated powders is observed. The mass loss was associated with exothermic and endothermic events occurring at 260-280 °C and 305 °C, respectively. The MW treatment results in a decrease of the exothermic peaks, which correspond to the crystallization of ceria and nickel hydroxide, respectively, indicating that the MW favors the crystallization of such phases.

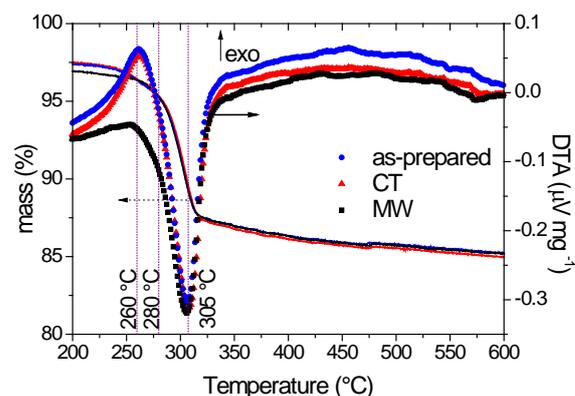


Fig. 1. TG/DTA of the as-prepared, CT, and MW powders of NiO/YSZ/CeO₂ composites.

Samples sintered at 1250 °C have apparent density corresponding to ~90% of the theoretical density, irrespectively of the thermal treatment. The Arrhenius plot of the total electrical conductivity of sintered pellets is shown in Fig. 2. The electrical conductivity values are similar for the three samples. Calculated activation energies (~380 meV) are in good agreement with previously reported values for NiO in this temperature range.

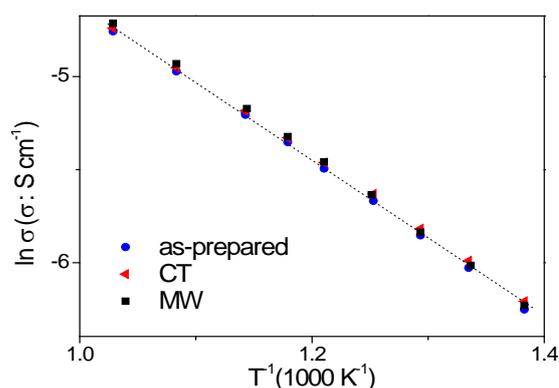


Fig. 2. Arrhenius plots of the total electrical conductivity of NiO/YSZ/CeO₂ composites.

Conclusions

The microwave-assisted hydrothermal treatment on coprecipitated NiO/YSZ/CeO₂ promotes the crystallization of nickel and ceria phases at low temperatures. However, such an effect has little influence on the electrical properties of sintered samples of the composite.

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