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Original Article

Laser powder bed fusion parameters to produce high-density Ti–53%Nb alloy using irregularly shaped powder from hydride-dehydride (HDH) process



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ABSTRACT

Laser powder bed fusion (LPBF) – also known as selective laser melting (SLM) – is a technology of additive manufacturing (AM) that offers benefits to the fabrication of implants. This approach can create customized and complex parts with low elastic modulus to reduce stress shielding. The use of irregularly shaped powder is not common due to its low flowability and low apparent density. However, its low cost arouses interest in the production of materials by this technology. This work discloses the processing window that allows the fabrication of Ti-53wt.%Nb alloy parts with high density using irregularly shaped powder from hydride–dehydride (HDH) process and analyzes the influence of the process parameters on the microstructure and hardness of the samples manufactured by LPBF. Energy densities (E_v) from 16 to 317 J/mm³ were investigated. Experimental density measurements by the Archimedes' principle and pore area fraction were calculated and relating to the density estimated based on X-ray diffraction (XRD) results of the HDH powder. Vickers hardness showed strong correlation to the content of interstitial elements of samples made under different E_v . This work proves that is possible to obtain samples with high density using HDH powder in LPBF and that the content of interstitial elements increased with the energy density, as well as the hardness of the alloy.

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1. Introduction

Laser powder bed fusion (LPBF) is a technology of additive manufacturing (AM) that uses one or more thermal sources to melt specific areas of powder particles that are spread in the form of layers across the built surface. Once the selected areas of one layer are molten, the build surface is lowered, more powder particles are spread, and new areas are molten. This process is repeated until the part is finished [1]. Detailed reviews of metallic components manufactured by this technology and their defects can be found in [2–5].

The usual feedstock material used in LPBF is spherical shaped powder produced by atomization due to its adequate flowability and apparent density (around 50%) that enable the fabrication of parts with high density [2]. Literature indicates that irregularly shaped powder, such as that fabricated by hydride–dehydride process (HDH), is not adequate to LPBF due to defects caused by inconsistent powder deposition as a consequence of its low flowability and apparent density [2,5–7]. Despite these factors, the lower cost of HDH powders [5,8,9] justifies further investigation of that route. Recently, Narra et al. [10] reported the obtaining of Ti–6Al–4V parts with relative density of 99.8% by electron beam powder bed fusion (EBPBF) from HDH powder after finding the adequate combination of processing parameters.

LPBF has already been used in the medical industry [11]. Currently, Ti–6Al–4V alloy is the most frequently used material in this application due to its high strength and biocompatibility [12,13]. However, aluminum (Al) and vanadium (V) have been claimed potentially cytotoxic. These elements can cause allergic reactions and possible diseases in the human body. There are reports that associate ions of Al and V to long-term health problems as peripheral neuropathy, osteomalacia, and Alzheimer's disease [14–16]. Human exposure to metal derivatives results in a wide range of specific toxicities that can lead to the diseases above and even cancer [17,18]. Another concern in this application is the bone resorption caused by stress shielding when the elastic modulus of the implant is much higher than the elastic modulus of the bone. The stress shielding is one of the most important causes of implant loss [19]. The stiffness mismatch between the bone and implant can be reduced by the insertion of pores in the implant [20–22], by using implant material with low elastic modulus, and/or by the control of its crystalline texture [23,24].

The Ti–53%Nb alloy is highly used in the fabrication of superconducting materials. It may be of interest to the medical industry due to its high biocompatibility and elastic modulus around 80 GPa [25–30]. Although this alloy presents these properties, so far there is only evidence of implant developments with the Ti–42%Nb alloy [31,32]. Moreover, it is known that niobium stabilizes the β phase of body-centered cubic (BCC) structure in the titanium-niobium system and it is possible to retain that phase in a metastable state in the Ti–54%Nb alloy [30,33–36]. The most recent paper about the lattice parameter of β phase for this composition was published by Pattanayak et al. [37], who found a value of 3.285 Å, conforming data of previous literature [33,38–40].

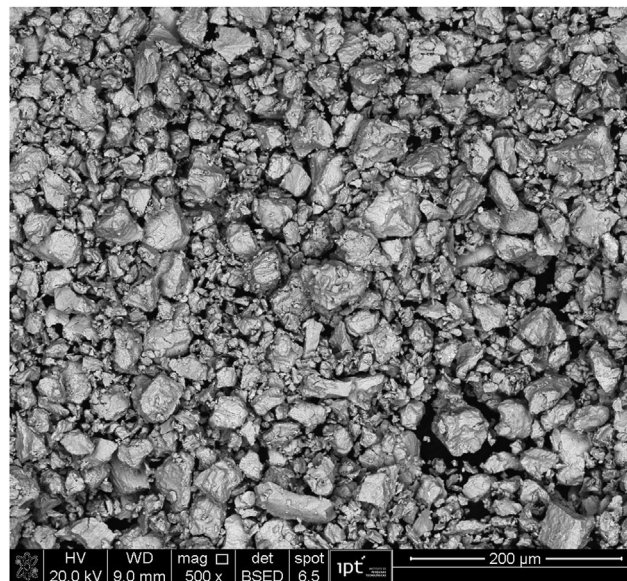


Fig. 1 – Irregular morphology of the powder manufactured by the HDH process (SEM).

In this paper, 10 mm cubes of the Ti–53%Nb alloy were manufactured by LPBF, using HDH powder. Some of them achieving low porosity depending on the power and scanning speed of the laser beam. Two methods were used to measure porosity, pore area fraction by metallography and relative density using the Archimedes' principle and specific mass of the alloy. To find the values of the relative density, the specific mass of the HDH powder was calculated by X-ray diffraction (XRD). Optical microscopy (OM), scanning electron microscopy (SEM), and energy-dispersive X-ray spectroscopy (EDX) were used to analyze and describe the microstructure and defects of the samples with high density. Finally, the hardness of the pieces and their content of interstitial elements were found and related. This work creates a path for the use of the Ti–53%Nb alloy in the fabrication of medical implants, demonstrating that it is possible to obtain material with high density from cheaper HDH powder when compared to the spherical shaped powder by LPBF. More details about that comparison are indicated in the work of Guzmán [41].

2. Experimental procedure

2.1. Powder characterization

Irregularly shaped powder was produced by hydride-dehydride process (HDH) from Ti–53%Nb alloy ingots, which were previously obtained by electron beam melting furnace (EBMF) at the Institute for Technological Research (IPT), in São Paulo, Brazil. The ingots of titanium (99.475% purity) and niobium (99.8% purity) were supplied by Acnis do Brasil and the Brazilian Metallurgy and Mining Company (CBMM), respectively. During the hydriding step, the alloy ingots were heated at 650 °C for 3 h, under a constant hydrogen pressure of 1.6 bar, followed by a controlled furnace cooling step. The hydride

alloy ingots were milled in a ring-type mill (Kawasaki Kobe, T-100) and sieved at $-53\ \mu\text{m}$. The dehydrating step of the alloys was carried out under high vacuum at $650\ ^\circ\text{C}$ for times up to 3 h, followed by a furnace cooling step under high vacuum. More information about similar procedures to those adopted in the obtention of HDH powders of the Ti–53%Nb alloy can be found in the relevant literature [42]. The powder morphology obtained by HDH process is shown in Fig. 1. The HDH powder had a particle size distribution described by Fig. 2, with figures of merit $12\ \mu\text{m}$ (D10), $31\ \mu\text{m}$ (D50), and $67\ \mu\text{m}$ (D90), measured by laser diffraction (Malvern, Mastersizer 2000). The irregularity of the particles was also evaluated by dynamic image analysis (Micromeritics, Particle Insight), showing mean values to circularity and Feret aspect ratio of 0.592 and 1.469, respectively. The circularity is a fractional measure, equal to 1 for a perfect circle. As for Feret aspect ratio, it is defined as the ratio between the Feret length and width, equal to 1 for perfect equiaxial symmetry. This explains the existence, in Fig. 1, of particles with size beyond $100\ \mu\text{m}$, albeit been screened by $53\ \mu\text{m}$ sieve. The powder presented an apparent density of $1.85\ \text{g/cm}^3$ with no flow in hall flowmeter funnel. The oxygen (O), nitrogen (N), and hydrogen (H) contents were determined by inert gas fusion method (LECO, ONH 836) and are listed in Table 1. Density was also measured in a helium gas pycnometer (AccuPyc II 1340).

2.2. Fabrication of the LPBF samples

In this study, 20 samples of Ti–53%Nb alloy in the form of cubes ($10 \times 10 \times 10\ \text{mm}^3$) were fabricated in a Concept Laser M2 printer installed at the SENAI Innovation Institute for Manufacturing Systems and Laser Processing (ISI), in Joinville, Brazil. The displacement of the build platform was chosen as $60\ \mu\text{m}$ to cope with the low flowability and low apparent density of the HDH powder. The spreading of the powder did not present special difficulties, even with the powder having a small fraction of larger particles. It should be taken into account that, with the relative apparent density of 30%, the steady state true layer thickness is $200\ \mu\text{m}$ (platform displacement/relative apparent density) [43]. The recoater

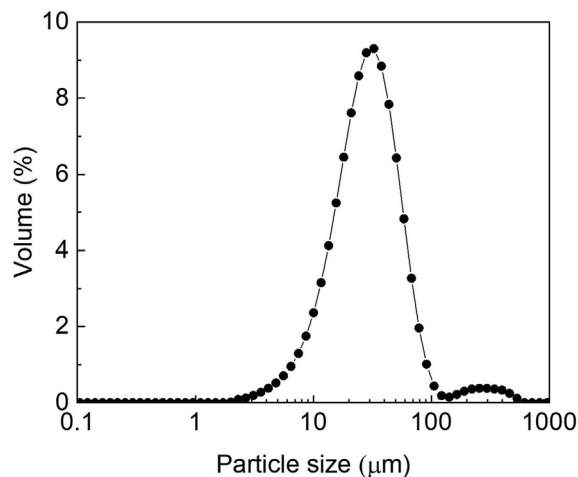


Fig. 2 – Particle size distribution of the Ti–53%Nb HDH powder.

Table 1 – Content of interstitial elements in the HDH powder.

Element	O	N	H
Content (wt.%)	0.34 ± 0.02	0.043 ± 0.005	0.0058 ± 0.003

speed was set at $65\ \text{mm/s}$, the same speed that was used in other experiments with spherical particles. The laser parameters for energy densities from 16 to $317\ \text{J/mm}^3$ are shown in Table 2. The hatch distance was kept constant at $105\ \mu\text{m}$. As the beam diameter is around $150\ \mu\text{m}$, the overlapping of melt pools can be estimated as 30%. The energy density (E_v) was calculated using the expression:

$$E_v \left(\frac{\text{J}}{\text{mm}^3} \right) = \frac{P_L (W)}{V_s \left(\frac{\text{mm}}{\text{s}} \right) H_s (mm) D_s (mm)}$$

where P_L is laser power, V_s is scanning speed, H_s is hatch distance and D_s is the layer thickness usually assumed in the literature, as the platform displacement [43].

The scanning strategy was a Chessboard tilted 67° relative to one of the walls, with islands of $5 \times 5\ \text{mm}^2$. Moreover, a variation of 90° of the laser scanning in each island was used for the subsequent layers as shown in Fig. 3. Furthermore, the process was carried out in a closed chamber continuously flushed with a shielding argon gas to keep the oxygen content to levels between 0.4 and 0.6%.

2.3. Density characterization

Estimates of the relative density were obtained using the Archimedes' principle and pore area fraction. Relative density and porosity are related so that their sum is 100%. The theoretical volumetric mass density was calculated based on the atomic proportion (37%Nb and 63%Ti), the lattice parameter and the niobium (Nb) and titanium (Ti) atomic mass (92.906 and $47.867\ \text{g/mol}$, respectively). To obtain the lattice parameter of the alloy, X-ray diffraction (XRD) was applied to the Ti–53%Nb HDH powder with the extrapolation method of Nelson-Riley [44], as described by the expression:

$$a_{ap} = a_{ext} + b \left(\frac{\cos^2 \theta}{\sin \theta} + \frac{\cos^2 \theta}{\theta} \right)$$

where a_{ext} is the best estimate of the lattice parameter, based on the extrapolation of the apparent lattice parameter calculated from each θ Bragg angle (a_{ap}). XRD tests were also conducted to check the existence of other phases in the microstructure of the LPBF samples produced with different energy densities. The analyses were done in a Philips-X'Pert PRO PW 3040/00 equipment at $45\ \text{kV}$ and $40\ \text{mA}$ with a Cu K_α radiation source.

To determine density by the Archimedes' principle, the initially printed cubes were ground with abrasive paper down to 1200 grit and paraffin wax was used to seal surface-connected porosity according to the ASTM standard B962-17 [45]; each sample was measured 3 times. On the other hand, for the pore area fraction method, the samples were polished with a mixture of colloidal silica and hydrogen peroxide until reaching a mirrored surface. Then, 9 micrographs of different zones in planes containing the building direction (BD) at $50\times$

Table 2 – Process parameters.

Sample	P_L (W)	V_S (mm/s)	E_V (J/mm ³)	Sample	P_L (W)	V_S (mm/s)	E_V (J/mm ³)
1	400	1000	63	11	200	200	159
2	400	800	79	12	200	400	79
3	400	600	106	13	200	600	53
4	400	400	159	14	200	800	40
5	400	200	317	15	200	1000	32
6	300	1000	48	16	100	200	79
7	300	800	60	17	100	400	40
8	300	600	79	18	100	600	26
9	300	400	119	19	100	800	20
10	300	200	238	20	100	1000	16

of magnification were used to measure the relative density using Fiji software [46].

2.4. Microstructure characterization of the LPBF samples

The samples with the highest relative density were chosen to have their microstructure and content of interstitial elements analyzed. After samples being polished, they were etched with Kroll’s reagent (100 ml of H₂O, 1–3 ml of HF, and 2–6 ml of HNO₃) to reveal their microstructure. Melt pools dimensions of depth and width were measured to observe the influence of the parameters. For this, micrographs were taken in a ZEISS Axio Observer.Z1m microscope and in a field emission gun Inspect 50 Scanning Electron Microscope (FEG-SEM) using secondary electrons (SE) and back-scattered electrons (BSE). Energy-dispersive X-ray spectroscopy (EDX) coupled to the SEM was used for taking chemical composition measurements.

Microhardness was measured in a DuraScan 70 machine using Vickers indentation with a load of 200 g. Each sample was measured on average 10 times in different locations for 15 s. Measurements of O, N, and H were done in samples manufactured with different energy density by inert gas fusion method.

3. Results and discussion

3.1. X-ray diffraction analysis

From the X-ray diffractograms shown in Fig. 4, only the diffraction peaks of β phase of BCC structure can be seen in the

HDH powder and samples 2 and 10 with energy densities of 79 and 238 J/mm³, respectively, as expected for this composition according to the literature [33,34,38]. The lattice parameter calculated by Nelson–Riley method approached 3.285 Å, confirming that found by Pattanayak et al. [37] and similar to that found by Baden and Weiss [40], which is the inorganic crystal structure data (ICSD) reference for the Ti–50%Nb composition. Based on that lattice parameter and assuming the composition did not change with processing, theoretical volumetric mass density was estimated as 6.046 g/cm³. In addition, the density value was also calculated by other methods. First, hydrostatic density (ρ_{Arq}) measurement of the Ti–53%Nb electron beam cast alloy (EBC) resulted in 6.042 g/cm³. Pycnometer density of the HDH powder resulted in 6.049 ± 0.002 g/cm³. Among those different estimates and based on the similarity of the lattice parameter found with the other authors [33,37–40], the value of 6.046 g/cm³ was chosen as the reference specific mass of the alloy to calculate the relative density of the LPBF samples. A recent paper about the Ti–53%Nb published by Mousavi et al. [47] used a density of 6 g/cm³, a round number that seems not accurate enough for AM applications.

3.2. Relative density and porosity

The density of each sample was determined by the Archimedes’ principle and pore area fraction method. The relative density was calculated in relation to the reference density value established as the specific mass of the alloy, 6.046 g/cm³. In Fig. 5, it is possible to see the influence of the variables, P_L and V_S , on the porosity. Samples 1, 2, 6, and 7 successfully

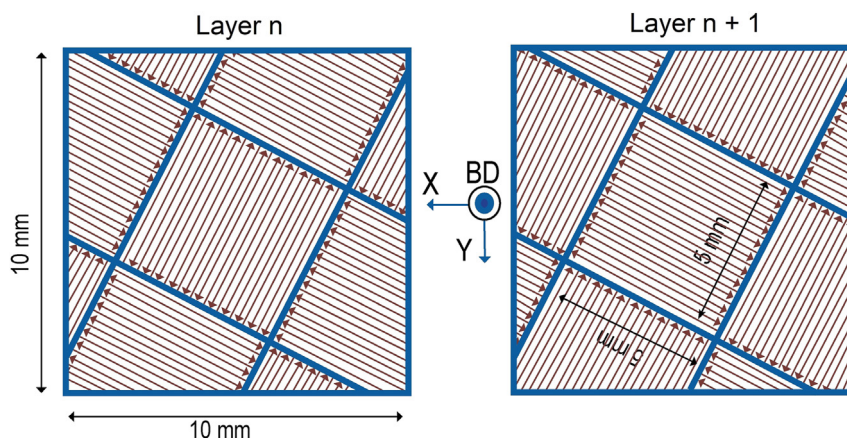


Fig. 3 – Scanning strategy. Layer n: Chessboard tilted 67° with islands of 5 × 5 mm². Layer n + 1: the same Chessboard with a variation of 90° of the laser scanning inside each island.

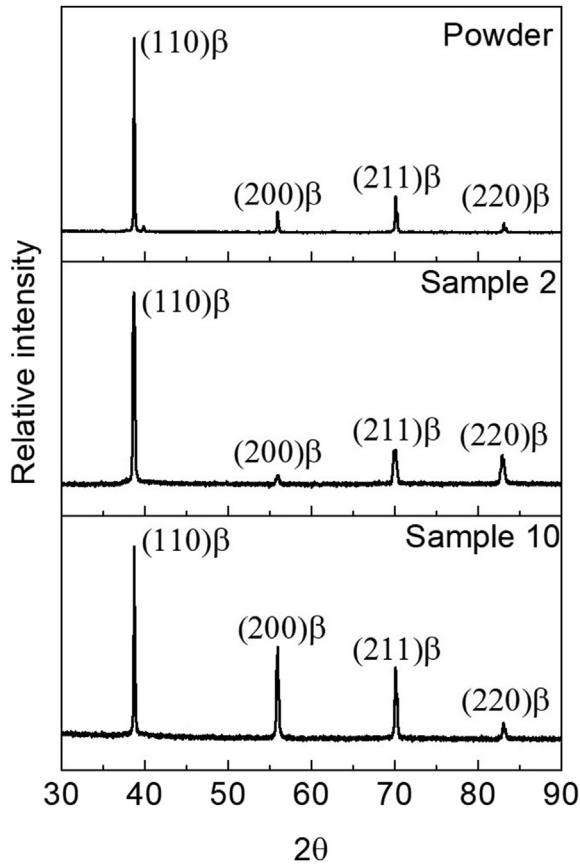


Fig. 4 – Diffractograms of the Ti–53%Nb HDH powder, sample 2 ($E_V = 79 \text{ J/mm}^3$) and sample 10 ($E_V = 238 \text{ J/mm}^3$) presenting the diffraction peaks of the β phase of BCC structure.

resulted in porosity values below 1%, as measured by both methods. The results by the Archimedes' principle and pore area fraction method differed significantly for samples produced with 100 and 200 W. This is justified because the polishing of samples with high pore area fraction is prone to enlarge the size of the pores. Nevertheless, its use carries special value for revealing the origin of the porosity defects

through their morphology, as will be discussed in the microstructural analysis.

The results of relative density obtained by Archimedes' principle as a function of the energy density are presented in Fig. 6. The chart shows that high laser power does not guarantee the highest density. The use of an excessive E_V decreases the density of the pieces and, in some cases, it does not allow the fabrication of samples due to the jamming of the spreading system [48], as it happened in samples 4 (159 J/mm^3) and 5 (317 J/mm^3). Results indicate that E_V is not a good parameter to predict sample density. While an E_V of 159 J/mm^3 to fabricate samples 4 and 11 led only to the fabrication of the latter, the use of an E_V of 79 J/mm^3 for samples 2, 8, 12, and 16 led to cubes with different values of relative density. Porosity depends mainly on power and scanning speed of the laser [5,48–50]. Therefore, it is better to analyze the values of relative density as in Fig. 5 because it presents a broader vision of the processing window used.

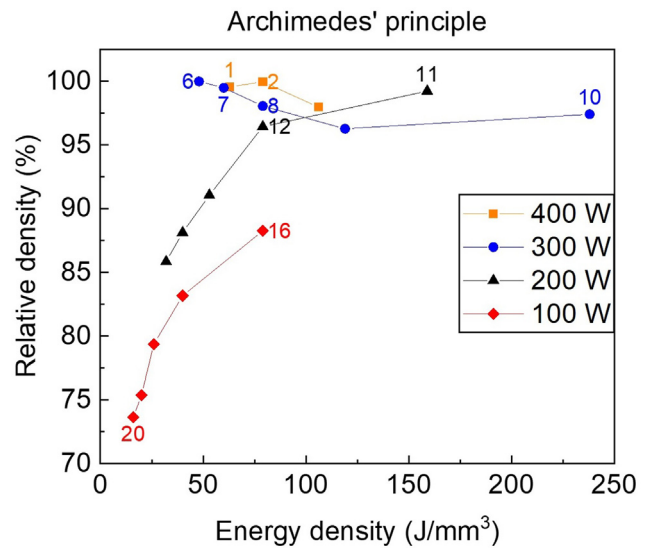


Fig. 6 – Relative density obtained by the Archimedes' principle as a function of energy density.

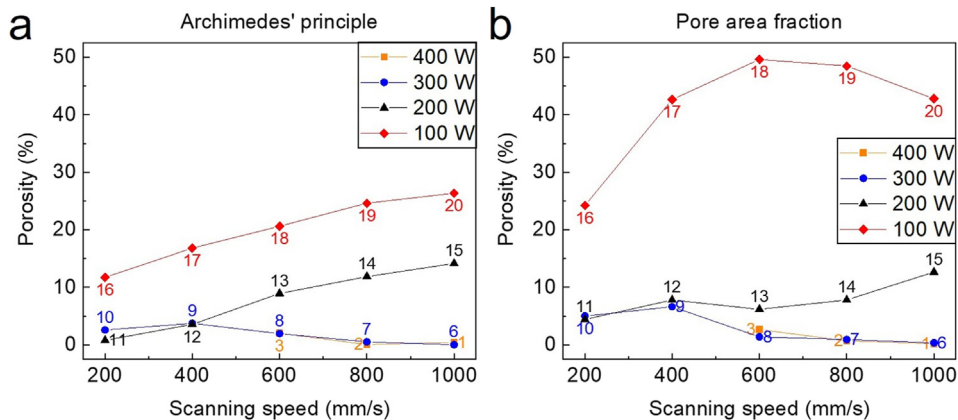


Fig. 5 – Porosity of the LPBF samples. a) Archimedes' principle and b) pore area fraction.

3.3. Microstructural analysis

The LPBF samples were classified according to their defects, using Gong et al. [48] classification in four zones: overheating, incomplete fusion, overmelting, and “fully dense”. In this paper, the “fully dense” zone is defined as the power-speed area where samples show porosity below 1% [48]. The micrographs of the samples are organized in a chart in Fig. 7 as a function of power and scanning speed of the laser. The fabrication of samples 4 and 5 had to be interrupted due to damage to the recoater by excessive heating and are classified as the overheating zone. They were manufactured with E_V of 159 and 317 J/mm³, respectively, and a high P_L of 400 W. The incomplete fusion and overmelting zones can be differentiated by the morphology of pores. In the incomplete melting zone, the porosity has irregular shape from the lack of fusion of some particles due to insufficient energy density [51]. As can be seen for laser power of 100 and 200 W, increasing laser speed led to increasing lack of fusion pores. In the overmelting zone, it is common to have spherical pores produced by trapped gas or vapor [52]. For laser power of 300 and 400 W, decreasing laser speed led to increasing the amount of spherical pores. The “fully dense” zone is formed by samples 1, 2, 6, and 7, produced with P_L from 300 to 400 W and V_S from 800 to 1000 mm/s. They showed a very small amount of pores, originated either from metal vapor, entrapped shielding gas, or even hydrogen from the HDH powder [5,52]. Therefore, that is the processing window to manufacture low porosity LPBF parts of Ti–53%Nb alloy from HDH powder.

Micrographs of the samples with the highest density (1, 2, 6, and 7) are shown in Fig. 8. Their microstructures are composed of melt pools and irregularly shaped grains in a β phase matrix. Melt pool boundaries can be seen due to segregation [53,54]. Moreover, grain boundaries crossing melt pool boundaries can be seen in samples 2 and 6 suggesting epitaxial growth.

It should be noted that the melt pools show different dimensions for each sample. The apparent depth and width of the melt pools are listed in Table 3, after measurements of samples in Fig. 8. The average depth is considerably larger than the platform displacement of 60 μ m, while the width is only 20% larger than the hatch distance (105 μ m). It was observed that samples fabricated with a P_L of 400 W presented greater depths than samples fabricated with a P_L of 300 W. These results suggest that the melt pool depth increases with P_L , in agreement with other authors [55,56], while the melt pool width does not vary considerably. Samples 1, 2, and 7 present goblet-shaped melt pools, whereas in sample 6 the melt pools show a trend to semicircle shape. In all of them the scanning strategy used resulted in overlapping between melt tracks and consecutive layers.

As expected by the high cooling rate and high temperature gradient in LPBF [2,57,58], all samples presented a cellular structure solidification, as exemplified by SEM images of samples 1 and 6, in Fig. 9. The darker cellular walls in the BSE images (Fig. 9 b, d, and f) indicate that those regions have a higher titanium content, as expected from the Ti–Nb phase diagram [35]. Some authors reported that cellular walls also include dislocation structures [49,57,59]. It is noted that there is a transition from coarse to fine cellular structure from the boundary to the center of the melt pools in all the micrographs, similar to the cell refinement described by Pham et al. [24]. SEM-EDX analyses were performed on the cellular structure in several regions. From the results listed in Table 4 and Fig. 9 for sample 6, it was possible to see that the segregation of Ti was higher at the cell walls. Nevertheless, in some regions of sample 1, it was not possible to find chemical differences between the cell walls and cell interior. This is because the cells are rods with diameters around 1 μ m, and that is the dimension of the interaction volume of SEM-EDX

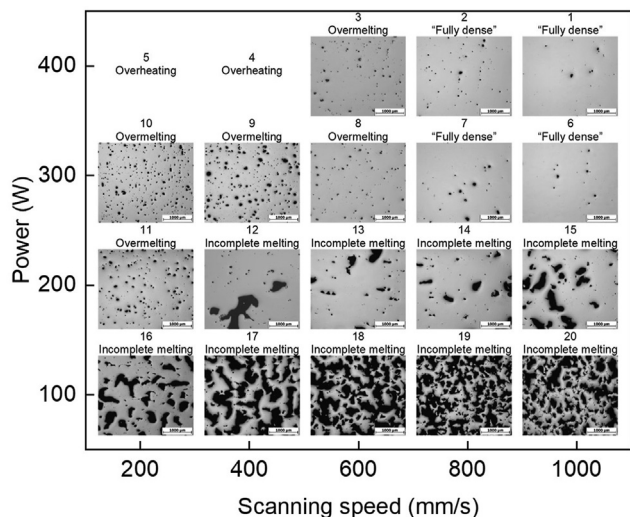


Fig. 7 – Micrographs of the LPBF Ti–53%Nb samples as a function of laser power and scanning speed. The scale bar is 1000 μ m and the vertical direction of the OM micrographs is parallel to the building direction.

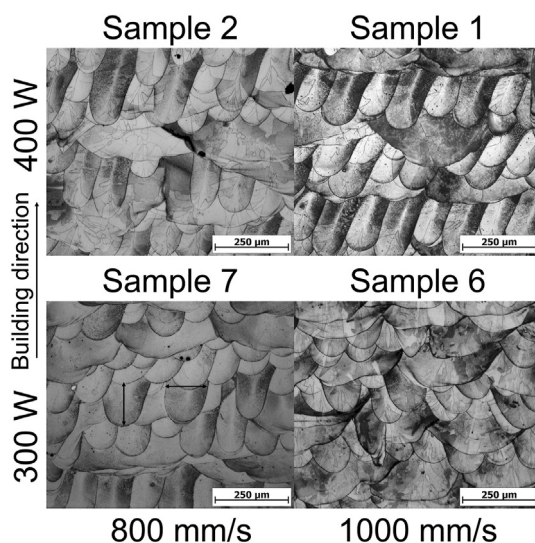


Fig. 8 – Microstructure of the high-density samples. Sample 1 $\rho_{Arq} = 99.57\%$, sample 2 $\rho_{Arq} = 99.97\%$, sample 6 $\rho_{Arq} = 99.99\%$, and sample 7 $\rho_{Arq} = 99.46\%$. Arrows in sample 7 indicate how the apparent melt pool dimensions were measured.

Table 3 – Apparent melt pool dimensions of “fully dense” samples in Fig. 8.

Sample	P _L (W)	V _S (mm/s)	Depth (μm)	Width (μm)
1	400	1000	170 ± 22	125 ± 16
2	400	800	186 ± 31	118 ± 11
6	300	1000	94 ± 25	128 ± 11
7	300	800	147 ± 33	124 ± 13

Table 4 – SEM-EDX analysis results taken in samples 1 and 6 in Fig. 9.

Sample	Position	Accelerating voltage (kV)	Element (wt.%)	
			Ti	Nb
1 (Fig. 9 b)	Cell wall	5.00	46	53
	Cell inner	5.00	46	53
6 (Fig. 9 d)	Cell wall	20.00	54	46
	Cell inner	20.00	48	52
6 (Fig. 9 f)	Cell wall	20.00	52	48
	Cell inner	20.00	50	50

analysis. Therefore, the sensibility of SEM-EDX is limited to perform segregation detection in this kind of samples.

3.4. Microhardness and interstitial element contents

The oxygen content of the HDH powder, shown in Table 1, is a major drawback of the HDH process [60]. Nevertheless, as the Ti–53%Nb alloy exhibits a BCC structure, the oxygen content may not impair ductility as much it does in other titanium alloys [61]. Ductility measurements will be presented in a future paper. The LPBF processing in a closed chamber with a shielding atmosphere where oxygen levels were controlled to a maximum of 0.6% led to a further increase in the oxygen and nitrogen content, as shown in Fig. 10 a. The effect of energy density on hardness is presented in Fig. 10 b. Energy density increased the absorption of interstitial elements from the

rather high content in the shielding atmosphere. It can be noted that the nitrogen increase was higher than the increase of oxygen content, in relative terms. Although the LPBF process strongly affects the grain size, texture, cellular structure, dislocation density, and segregation [59,62], literature has shown that both, oxygen and nitrogen, harden titanium alloys, being nitrogen the strongest hardener [63,64]. There is a significant increase in hardness, from the ingot (EBC) (160 HV_{0.2}) to the LPBF samples, but it must be considered that the oxygen content increased from 0.073 ± 0.001%, in the as cast condition, to 0.34 ± 0.02% in the HDH powder and then to the values shown in Fig. 10 a. The nitrogen content changed from

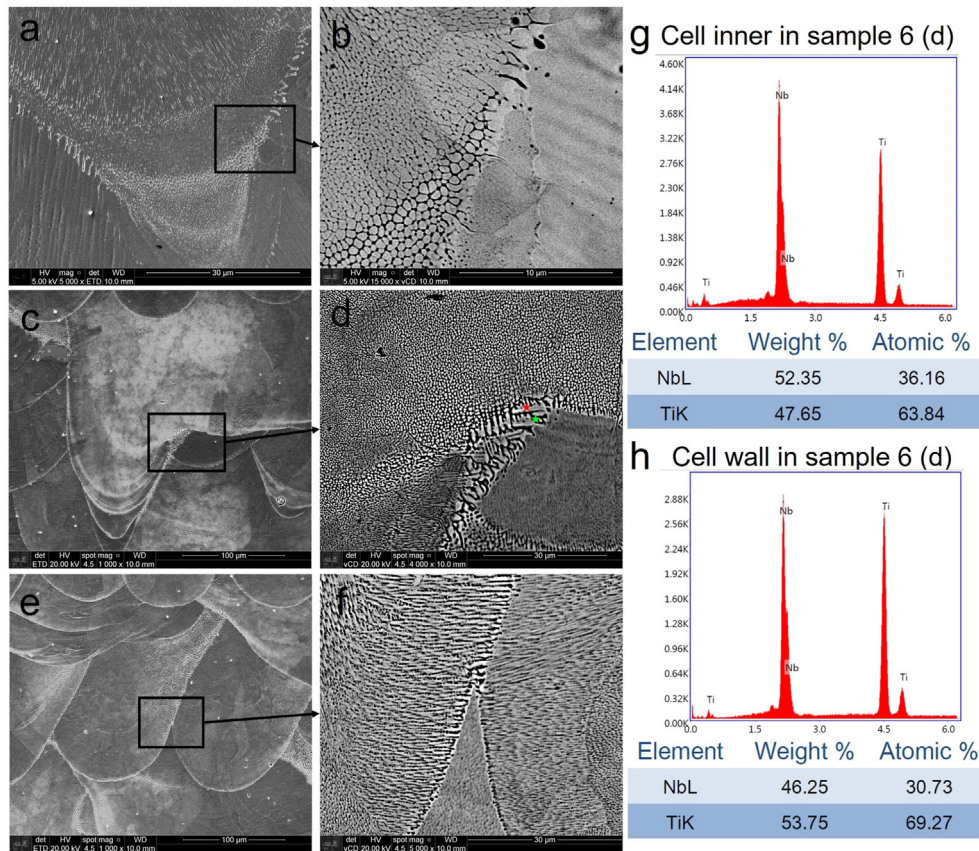


Fig. 9 – Cellular structure in sample 1 and 6. a) and b) Sample 1 (P_L = 400 W and V_S = 1000 mm/s) and c) to f) sample 6 (P_L = 300 W and V_S = 1000 mm/s). a), c) and e) SEM-SE and b), d) and f) SEM-BSE. g) and h) SEM-EDX of sample 6 in red and green points in d). (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

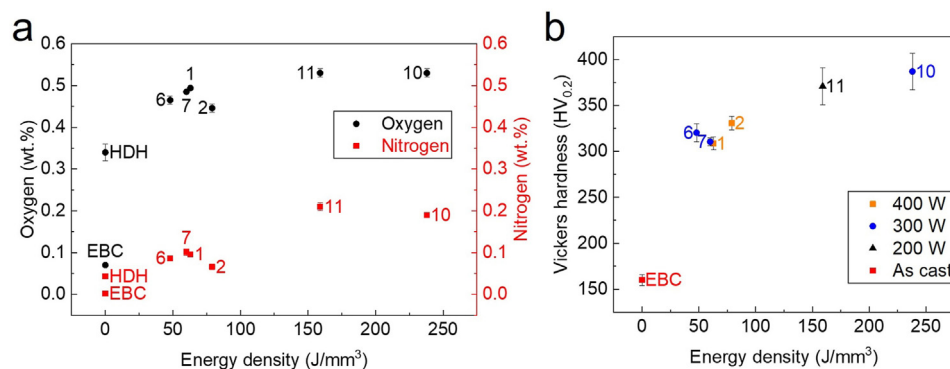


Fig. 10 – Effect of energy density on a) oxygen and nitrogen contents and b) Vickers hardness of the samples.

0.002 ± 0.001% to 0.043 ± 0.005% and to 0.2 ± 0.01%, respectively. Therefore, the increase in the hardness of the LPBF samples can be attributed to the absorption of interstitial atoms, that are solid solution strengtheners. The content of hydrogen increased with respect to the initial powder (0.0058 ± 0.003 wt.%) and kept stable after reaching 0.02% for all the manufactured samples. It needs to be said, that the content of H was very low when compared with the contents of O and N reached by the LPBF samples, for that reason, H is not presented in Fig. 10 a.

These results show that if high laser power needs to be used, a reduction in the levels of oxygen in the closed chamber is necessary. Therefore, further studies are needed to reduce the oxygen content in the closed chamber aiming at lowering the hardness of the LPBF Ti–53%Nb alloy from HDH powder with processing parameters outside from the overheating zone. These analyses can be performed using more combinations of power and scanning speed of the laser inside the “fully dense” zone found in order to obtain specific microstructures (different melt pool and grain shapes).

4. Conclusions

Parts of Ti–53%Nb alloy from irregularly shaped hydride-dehydride (HDH) powder were manufactured by laser powder bed fusion (LPBF) to find a high-density processing window. The conclusions can be summarized as follows:

- It was possible to obtain samples with relative density higher than 99%, assuming a volumetric mass density of 6.046 g/cm³ determined from a lattice parameter of 3.285 Å for the Ti–53%Nb alloy. The high-density process window was found in the laser power interval of 300–400 W and scanning speeds higher than 800 mm/s.
- The final oxygen content is near 0.5%, as the HDH powder had an oxygen content of 0.34%, and the chamber atmosphere oxygen content was kept only below 0.6%. Further studies in the processing parameter range to avoid overheating and absorption of interstitial elements are necessary.
- The microstructure composed by β phase of BCC structure presented irregularly shaped grains, cellular structure with segregation of Ti at the cell walls, and different melt pool

forms. Moreover, the oxygen content in the samples with high relative density, below 0.5%, did not affect the microstructure. However, it is recommended to decrease it for medical applications.

- Samples manufactured with high energy density were more likely to absorb oxygen, and their hardness was affected by the content of interstitial elements in the samples. The values varied from 309 to 387 HV_{0.2}.

Data availability statement

All data reported in this paper is contained within the manuscript.

Declaration of Competing Interest

The authors declare that they have no conflict of interest.

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